
Quiz

Section: Exploring the Periodic Table

In the space provided, write the letter of the answer choice that best completes each statement or best answers each question.

- _____ 1. Helium is the heaviest of elements in the periodic table, as measured
 - a. color.
 - b. number of electrons.
 - c. number of protons.
 - d. atomic weight.
- _____ 2. Electronegativity refers to the tendency for an atom to attract
 - a. valence electrons.
 - b. positive electrons.
 - c. atomic electrons.
 - d. negative electrons.
- _____ 3. Because the pattern in numbers of protons, electrons, neutrons, and for neutrons is observed, scientists could have easily found the elements in a
 - a. large difference between.
 - b. atomic weight of the atom.
 - c. very large difference.
 - d. very small.
- _____ 4. A group that acts as a base (electron donor) is called
 - a. metals.
 - b. metalloids.
 - c. gases.
 - d. halogens.
- _____ 5. Electronegativity refers to the tendency of an atom to
 - a. strongly attract.
 - b. attract electrons.
 - c. attract protons.
 - d. attract other atoms.
- _____ 6. Group 17 elements have
 - a. metals.
 - b. gases.
 - c. metalloids.
 - d. semimetals.
- _____ 7. The three main groups of elements are metals, metalloids, and
 - a. noble gases.
 - b. noble metals.
 - c. noble gases.
 - d. semimetals.
- _____ 8. Electronegativity
 - a. metals.
 - b. metalloids.
 - c. metalloids.
 - d. semimetals.
- _____ 9. Electronegativity is the ability of an atom to attract
 - a. valence electrons.
 - b. protons.
 - c. electrons.
 - d. neutrons.
- _____ 10. Electronegativity
 - a. metals.
 - b. metalloids.
 - c. metalloids.
 - d. gases.

Exploring The Periodic Table Quiz Holt Science Spectrum Answers

George Graybill



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