

Thinking Time

Explain to students that a continuous spectrum of light can be split into a continuous range of

discrete spectral lines

(discrete spectral lines observed)

Bohr's idea

(Bohr's idea)

Bohr's idea is that

electrons follow discrete orbits around protons that are stationary and that the electrons can only move between orbits. When a transition occurs between two stationary, higher energy orbits, energy is emitted as light.



Bohr's model



Bohr's Atomic Model

Bohr's Atomic Model is a model of the atom.

It is a model of the atom that is based on the idea that

electrons move in discrete orbits around the nucleus.

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Exploring Gas Laws Phet Answers

H Kauffman



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