

No

Activity A: Balancing equations

Get the Gizmo ready:

- Check that the **Synthesis** reaction is selected and that all **coefficients** are set to one. (The coefficients are the numbers in the boxes.)



Introduction: The equation $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ is unbalanced because there are two oxygen atoms on the reactants side of the equation, and only one on the products side of the equation. To balance the equation, you cannot change the structure of any of the molecules, but you can change the number of molecules that are used.

Question: How are chemical equations balanced?

1. **Balance:** Turn on **Show histograms**. The equation is balanced when there are equal numbers of each type of atom represented on each side of the equation. In the Gizmo, use the up and down arrows to adjust the numbers of hydrogen, oxygen, and water molecules until the equation is balanced. When you are done, turn on **Show summary** to check your answer.

Write the balanced equation here: $2 \text{H}_2 + 1 \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$

2. **Explore:** Turn off **Show summary**. Use the Gizmo's reaction drop-down menu to use other equations, and balance them. Check your answers and then write the balanced equations.



3. **Challenge:** Balance the following chemical equations. (These equations are not in the Gizmo.)



This work is derived from Balancing Chemical Equations Gizmo, Version 1.0, © 2005 by Science Education Resource Project (SERP).

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Explore Learning Chemical Equations Answers

R Barnett



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