Test bank chapter (9)

Choose the most correct answer

The two types of chemical bonds commonly found in compounds are:

- a) doric and covalent.
- b) ionic and electrolytic.
- c) ionic and covalent.
- d) electrolytic and compound.

2. The electrons used by atoms to form chemical bonds are the:

- a) core electrons.
- b) valence electrons.
- c) lone pair electrons.
- d) unpaired electrons.

3. "Atoms tend to gain, lose, or share electrons until they are surrounded by eight valence electrons" is a statement of:

- a) the rule of octaves.
- b) the double quartet rule.
- c) the eight electron rule.
- d) the octet rule.

4. When a transition metal atom becomes a +1 ion, the electron lost usually comes from what type of orbital?

- b) f
- ch d.
- di s

A molecule of CS₃ contains.

- a) two single bonds.
- b) two double bonds.
- c) one single bond and one double bond.
- d) one single bond and one triple bond.

6. An atom in the ground state has atomic number Z=5. Choose the correct electron-dot structure which represents this atom? ANS. B

- (B) ₹ -
- (C) +X -
- (D) =5_-

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