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web sep 24 2020 ans 3 6 the cell in which the following reaction occurs $2\text{Fe}^{3+}(\text{aq}) + 2\text{I}^{-}(\text{aq}) \rightarrow 2\text{Fe}^{2+}(\text{aq}) + \text{I}_2(\text{s})$ has e cell 0 236 V at 298 K calculate the standard gibbs

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web reason when successive electrons are added to the orbitals in the same principal quantum level the shielding effect of inner core of electrons does not increase very much to

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