lonic Bonding Practice 2

Instructions: Show how the following elements will ionically band using the steps below:

- Draw the lewis dot structure using the number of valence electrons for each element.
 - Note: You might want to use different colored pencits or data and x's for different elements to help remember which electrons belong to which element.
- Move the electrons and draw grows to show the transfer of electrons to fulfill the rester rule.
- Write the oxidation number for each ion based on how many negatively charged electrons have been alven away or accepted.
- Continue to add atoms until every atom has fulfilled the acter rule.
- Double check and make certain all axidation numbers add up to zero.
- Write the chemical formula by writing the number of each element in subscript.
- Write the name of the element. If the element has accepted electrons (unions)
 make the last three letters "ide".

Na Na Page 11-1	Li Ci
Chemical Name Sodium Fluoride	Chemical Formula:
3. Magnesium and Ovygen:	4. Lithaum and Phosphorus
Chemical Formula:	Chemical Formula:

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Ian Pickup

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Whispering the Techniques of Language: An Psychological Quest through Ionic Bonding Practice Worksheet Filetype

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