

Ionic and Covalent Bonds Study Guide

Be able to define the following terms:

- | | | |
|-------------------|-----------------|--------------------|
| ➤ covalent bond | ➤ anion | ➤ polyatomic ion |
| ➤ Lewis structure | ➤ cation | ➤ hydrogen bonding |
| ➤ molecule | ➤ chemical bond | ➤ valence electron |
| ➤ polar bond | ➤ ionic bond | |
| ➤ non-polar bond | ➤ ion | |

1. A group of covalently bonded atoms that acts together as one charged atom is a polyatomic ion.
2. A chemical bond that occurs when atoms share electrons is a(n) covalent bond.
3. The octet rule tells you how many electrons an atom must gain, lose, or share to become stable.
4. The oxidation number of an atom is shown with a roman numeral in parenthesis.
5. How many electrons are needed in the outer energy levels of most atoms for the atom to be chemically stable? **8 (Eight)**
6. Why do the noble gases NOT form compounds readily? **They have 8 electrons in their outer electron shell.**
7. What is the total number of atoms in the compound $\text{Ca}(\text{ClO}_3)_2$? **9 (Nine)**
8. Explain why it is correct to say that a compound like calcium carbonate, CaCO_3 , contains both ionic and covalent bonds. **Carbonate is a polyatomic ion (covalent bonded) and then it forms an ionic bond with Calcium.**
9. Name or write the formulas for the following ionic compounds:
 - a. Al_2O_3 **Aluminum Oxide**
 - b. NH_4Cl **Ammonium Chloride**
 - c. FeO **Iron (II) Oxide**
 - d. manganese (V) oxide **Mn_2O_5**
 - e. calcium chloride **CaCl_2**
 - f. lead (IV) nitrate **$\text{Pb}(\text{NO}_3)_4$**

Ionic And Covalent Bonding Study Guide

IM Harris



Ionic And Covalent Bonding Study Guide:

Ionic And Covalent Bonding Study Guide Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Ionic And Covalent Bonding Study Guide**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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