

## Balancing Chemical Equations Practice

**Step 1** – Take Inventory of the elements and atoms on the product and reactant side.

**Step 2** – Is it balanced?

**Step 3** – If unbalanced, change coefficients until it's balanced.



$$\text{C} - 1 = \text{C} - 1$$

$$\text{H} - 4 \neq \text{H} - 4$$

$$\text{Cl} - 8 \neq \text{Cl} - 8$$

**Balanced!**

# How To Balance Chemical Equations Study Guide

**Yijin Wang**



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