

Assessment 1**Chapter Test B****Chapter: Chemical Equations and Reactions**

Part 1 In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The precipitation of a slightly soluble solid compound from a solution of two ionic reactants is called the formation of a
a. gas.
b. precipitate.
c. combustion reaction.
d. halogen.
- _____ 2. To balance a chemical equation, it is permissible to adjust the
a. coefficients.
b. subscripts.
c. formulae of the products.
d. number of products.
- _____ 3. In a chemical equation, the symbol (aq) indicates that the substance is
a. melted.
b. dissolved in water.
c. an acid.
d. insoluble.
- _____ 4. The tendency for a replacement reaction to occur depends on the
a. relative reactivities of the elements in the activity series of metals.
b. temperature difference.
c. relative electronegativity values.
d. relative reactivities of the elements in the activity series of metals.
- _____ 5. The coefficients in a chemical equation
a. indicate the number of moles of each substance that react.
b. show the number of grams of each substance that react.
c. are the molar masses of the substances.
d. show the relative electronegativity values.
- _____ 6. If metal X is more reactive than metal Y in the activity series, then
a. X will replace ions of Y in solution.
b. Y will replace ions of X in solution.
c. Y will react with cold water.
d. X will react with cold water.

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Holt Chemistry Holt Rinehart & Winston, 2003-01-24 Modern Protein Chemistry Gary C. Howard, William E. Brown, 2001-09-12 In recent years interest in proteins has surged This resurgence has been driven by the expansion of the post genomic era when structural genomics and proteomics require new techniques in protein chemistry and new applications of older techniques Protein chemistry methods are used by nearly every discipline of biomedical research Many techniques *Studies on proteins from the Carlsberg laboratory* Soren Peter Lauritz Sørensen, 1917 Food and Nutrition Information and Educational Materials Center Catalog Food and Nutrition Information Center (U.S.), 1976 **Catalog.**

Supplement - Food and Nutrition Information and Educational Materials Center Food and Nutrition Information and Educational Materials Center (U.S.), **Application of Solution Protein Chemistry to Biotechnology** Roger L. Lundblad, 2009-05-12 Reflecting the versatility of the author's science and the depth of his experience Application of Solution Protein Chemistry to Biotechnology explores key contributions that protein scientists can make in the development of products that are both important and commercially viable and provides them with tools and information required for successfu **Catalog** Food and Nutrition Information Center (U.S.), 1973 **Proteome Informatics** Conrad Bessant, 2016-11-23 Bringing together world experts to provide clear explanations of the key algorithms workflows and analysis frameworks Proteome Informatics will provide a detailed introduction to the main informatics topics that underpin the various LC MS MS protocols used for protein identification and quantitation Biophysical Chemistry of Proteins Engelbert Buxbaum, 2025-07-10 This textbook designed for all scientists interested in protein research provides a thorough overview of laboratory methods for the biophysical chemistry of proteins This new edition completely restructured and expanded for ease of learning includes sections on analytical techniques working with proteins protein size and shape protein structure enzyme kinetics industry enzymology and a new section on special statistics **American Druggist**, 1925

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