

- Q.20 For a reversible reaction forward and backward rate constant respectively are 2.5×10^{-4} and 7.5×10^{-4} . What is the value of equilibrium constant [RPMT-2001]
 (1) 3 (2) $1/3$ (3) 1 (4) None
- Q.21 If the equilibrium constant for the reaction $N_2 + 3H_2 \rightarrow 2NH_3$ is K then the equilibrium constant for the reaction $2N_2 + 6H_2 \rightarrow 4NH_3$ would be equal to [RPMT-2002]
 (1) K^2 (2) \sqrt{K} (3) $\frac{1}{\sqrt{K}}$ (4) $\frac{1}{K^2}$
- Q.22 The reaction quotient (Q) for the reaction $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$ is given by $Q = \frac{[NH_3]^2}{[N_2][H_2]^3}$. The reaction will proceed from right to left if [CBSE-PMT-2003]
 (1) $Q < K_e$ (2) $Q > K_e$ (3) $Q = 0$ (4) $Q = K_e$
- Q.23 The following equilibria are given [CBSE-PMT-2003]
 $N_2 + 3H_2 \rightarrow 2NH_3$ K_1
 $N_2 + O_2 \rightarrow 2NO$ K_2
 $H_2 + \frac{1}{2}O_2 \rightarrow H_2O$ K_3
 The equilibrium constant of the reaction $2NH_3 + \frac{5}{2}O_2 \rightarrow 2NO + 3H_2O$, in terms of K_1 , K_2 and K_3 is :
 (1) $K_1 K_2 / K_3$ (2) $K_1 K_3^3 / K_2$ (3) $K_2 K_3^3 / K_1$ (4) $K_1 K_2 K_3$
- Q.24 For the gas phase reaction,
 $C_2H_4 + H_2 \rightleftharpoons C_2H_6$ ($\Delta H = -32.7$ kcal),
 carried out in a vessel, the equilibrium conc. of C_2H_4 can be increased by [AMU 2003]
 (1) Decreasing the pressure (2) Increasing the temperature
 (3) Removing some C_2H_6 (4) Adding some H_2
- Q.25 Consider the following reversible reaction at equilibrium, $2H_2O(g) \rightleftharpoons 2H_2(g) + O_2(g)$
 $\Delta H = 24.7$ kJ. Which one of the following changes in conditions will lead to maximum decomposition of $H_2O(g)$? [Kerala CEE 2004]
 (1) Increasing both temperature and pressure
 (2) Decreasing temperature and increasing pressure
 (3) Increasing temperature and decreasing pressure
 (4) Increasing temperature at constant pressure
- Q.26 In the reaction, $A_2(g) + 4B_2(g) \rightleftharpoons 2AB_4(g)$ $\Delta H < 0$, the formation of $AB_4(g)$ will be favoured at [MP PMT 2004]
 (1) High temperature, low pressure (2) Low temperature, high pressure
 (3) High temperature, high pressure (4) Low temperature, low pressure

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Environmental Chemistry Mcqs With Answers: Bestsellers in 2023 The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the fascinating narratives that have captivated audiences this year. The Must-Read : Colleen Hoover "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Environmental Chemistry Mcqs With Answers : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Environmental Chemistry Mcqs With Answers : Delia Owens "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a masterful and gripping novel that will keep you guessing until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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