# Naming Ionic Compounds - Answer Key

Give the name of the following ionic compounds:

#### Name

1)	Na <sub>2</sub> CO <sub>3</sub>	sodium carbonate
2)	NaOH	sodium hydroxide
3)	MgBr <sub>2</sub>	magnesium bromide
4)	KCI	potassium chloride
5)	FeCl <sub>2</sub>	iron (II) chloride
6)	FeCl <sub>3</sub>	iron (III) chloride
7)	Zn(OH) <sub>2</sub>	zinc hydroxide
8)	Be <sub>2</sub> SO <sub>4</sub>	beryllium sulfate
9)	CrF <sub>2</sub>	chromium (II) fluoride
10)	Al <sub>2</sub> S <sub>3</sub>	aluminum sulfide
11)	PbO	lead (II) oxide
12)	Li <sub>3</sub> PO <sub>4</sub>	lithium phosphate
13)	Titl4	titanium (IV) iodide
14)	Co <sub>3</sub> N <sub>2</sub>	cobalt (II) nitride
15)	Mg <sub>3</sub> P <sub>2</sub>	magnesium phosphide
16)	Ga(NO <sub>2</sub> ) <sub>3</sub>	gallium nitrite
17)	Ag <sub>2</sub> SO <sub>3</sub>	silver sulfite
18)	NH <sub>4</sub> OH	ammonium hydroxide
19)	Al(CN) <sub>3</sub>	aluminum cyanide
20)	Be(CH <sub>3</sub> CO	O) <sub>2</sub> beryllium acetate

# **Elements And Bonding Worksheet Chemfiesta**

**Lynn Van Gorp** 

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web kw the product of the concentrations of the hydrogen ions and hydroxide ions acidic solution one which h is greater than the oh basic solution one in which the h is less than the oh alkaline solutions the h of a basic solution is less than  $1 \times 10 \times 7 = 100$  m they are the same as basic solutions

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web conjugate acid base pair two substances related by the loss or gain of a single hydrogen ion hydronium ion h3o a water molecule that gains a hydrogen ion becomes this which is positively charged what did lewis propose an acid accepts a pair of electrons during a reaction while a base donates a pair of electrons

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