Teacher Guide: Element Builder



Learning Objectives

Students will...

- Compare the sizes, charges, and relative positions of the subatomic particles: protons, neutrons, and electrons.
- Relate an atom's number of protons to its charge, name, and atomic number.
- Calculate the mass number of an atom by summing the protons and neutrons.
- Understand the definition of isotope.
- Observe how electrons fill the first three energy levels.
- Create an electron dot diagram for each of the first 20 elements.



Vocabulary

atom, atomic number, electron, electron dot diagram, element, energy level, ion, isotope, mass number, neutron, nucleus, periodic table, proton, radioactive, valence electrons

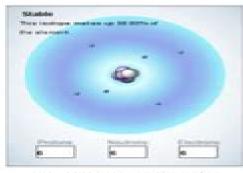


Lesson Overview

The Element Builder Gizmo™ allows students to construct atoms by adding protons, neutrons, and electrons. As they add particles, students can see the atomic number, mass number, charge, and electron dot diagram of the element.

The Student Exploration sheet contains two activities and an Extension:

- Activity A Students determine the mass number and atomic number of an atom.
- Activity B Students observe electron arrangements and create electron dot diagrams.
- Extension Students see how valence electrons relate to the periodic table of elements.



Model of a carbon atom



Suggested Lesson Sequence

1. Pre-Gizmo activity

(1 − 2 weeks)

Atoms are too tiny to see, yet much about their structure had been deduced by the early part of the twentieth century. The development of atomic theory is one of the central narratives of physics and well worth learning more about.

Divide the class into groups and assign to each group one of the critical scientists involved in atomic theory. These could include Democritus, John Dalton, Dmitri Mendeleev, J.J. Thomson, Robert Millikan, Ernest Rutherford, Marie Curie, James Chadwick, Niels Bohr, Enrico Fermi, and others (see Selected Web Resources).

After researching their scientist via the Internet, students can present their findings with posters, atom models, and even reenactments of famous experiments.



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Table of Contents Element Builder Gizmo Teachers Guide

- 1. Understanding the eBook Element Builder Gizmo Teachers Guide
 - The Rise of Digital Reading Element Builder Gizmo Teachers Guide
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Element Builder Gizmo Teachers Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Element Builder Gizmo Teachers Guide
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Element Builder Gizmo Teachers Guide
 - Personalized Recommendations
 - Element Builder Gizmo Teachers Guide User Reviews and Ratings
 - Element Builder Gizmo Teachers Guide and Bestseller Lists

- 5. Accessing Element Builder Gizmo Teachers Guide Free and Paid eBooks
 - Element Builder Gizmo Teachers Guide Public Domain eBooks
 - Element Builder Gizmo Teachers Guide eBook Subscription Services
 - Element Builder Gizmo Teachers Guide Budget-Friendly Options
- 6. Navigating Element Builder Gizmo Teachers Guide eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Element Builder Gizmo Teachers Guide Compatibility with Devices
 - Element Builder Gizmo Teachers Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Element Builder Gizmo Teachers Guide
 - Highlighting and Note-Taking Element Builder Gizmo Teachers Guide
 - Interactive Elements Element Builder Gizmo Teachers Guide
- 8. Staying Engaged with Element Builder Gizmo Teachers Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Element Builder Gizmo Teachers Guide
- 9. Balancing eBooks and Physical Books Element Builder Gizmo Teachers Guide
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Element Builder Gizmo Teachers Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Element Builder Gizmo Teachers Guide
 - Setting Reading Goals Element Builder Gizmo Teachers Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Element Builder Gizmo Teachers Guide
 - Fact-Checking eBook Content of Element Builder Gizmo Teachers Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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