

Chapter 1

Introduction to Electronics

Question A.1. Atomic Structure

- (a) The atom is the smallest constituent of matter that retains all the chemical properties of the element.
- (b) The atom consists of three subatomic particles, viz. electron, proton and neutron.

Question A.2. Materials Used for Electronics

- (a) The materials are generally classified into four main categories: (i) conductors, (ii) semiconductors, (iii) insulators and (iv) superconductors.
- (b) The atoms containing valence electrons are called conductors.
- (c) An atom having a partially filled valence shell is called a semiconductor.

Question A.3. Current in Semiconductors

- (a) Movement of electrons from one atom to another is called drift current.
- (b) Current is produced when there is a net movement of electrons in a material.

Question A.4. N-Type and P-Type Semiconductors

- (a) In an N-type semiconductor, the valence electrons are replaced by free electrons. This is done by adding a small amount of pentavalent impurity to the material. The impurity atoms have five valence electrons, while the semiconductor material has four.
- (b) In a P-type semiconductor, the valence electrons are replaced by free holes. This is done by adding a small amount of trivalent impurity to the material. The impurity atoms have three valence electrons, while the semiconductor material has four.

Question A.5. The PN Junction

- (a) The PN junction is formed by joining an N-type semiconductor with a P-type semiconductor. The junction is formed by the diffusion of electrons from the N-type material to the P-type material. This results in the formation of a depletion region, which is a region where the concentration of free electrons and holes is zero.
- (b) The PN junction is a semiconductor device that is used to convert electrical energy into light energy. It is used in a variety of applications, including light-emitting diodes (LEDs), solar cells, and photodiodes.

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Arthur James Wells



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