

- Which electron configuration represents an atom in an excited state?  
A)  $1s^2 2s^2 2p^6 3p^1$   
B)  $1s^2 2s^2 2p^6 3s^2 3p^1$   
C)  $1s^2 2s^2 2p^6 3s^2 3p^2$   
D)  $1s^2 2s^2 2p^6 3s^2$
- All of the elements in Period 3 have a total of 2 electrons in the  
A) 2s sublevel  
B) 3s sublevel  
C) 2p sublevel  
D) 3p sublevel
- Which atom in the ground state has three unpaired electrons in its outermost principal energy level?  
A) Li  
B) B  
C) N  
D) Ne
- What is the total number of valence electrons in an atom with the electron configuration  $1s^2 2s^2 2p^6 3s^2 3p^3$ ?  
A) 6  
B) 2  
C) 3  
D) 5
- Which orbital notation represents an atom of beryllium in the ground state?  
A) 

1s	2s	2p		
<div style="text-align: center;">↑</div>	<div style="text-align: center;">↑</div>	<div style="text-align: center;">↑</div>	<div style="text-align: center;">↑</div>	<div style="text-align: center;"> </div>

  
B) 

1s	2s	2p		
<div style="text-align: center;">↑↓</div>	<div style="text-align: center;">↑</div>	<div style="text-align: center;">↑</div>	<div style="text-align: center;"> </div>	<div style="text-align: center;"> </div>

  
C) 

1s	2s	2p		
<div style="text-align: center;">↑↓</div>	<div style="text-align: center;">↑↓</div>	<div style="text-align: center;"> </div>	<div style="text-align: center;"> </div>	<div style="text-align: center;"> </div>

  
D) 

1s	2s	2p		
<div style="text-align: center;">↑</div>	<div style="text-align: center;"> </div>	<div style="text-align: center;">↑</div>	<div style="text-align: center;">↑</div>	<div style="text-align: center;">↑</div>
- Which element has a completely filled third principal energy level?  
A) Ar  
B) N  
C) Fe  
D) Zn
- Which atom in the ground state has only 3 electrons in the 3p sublevel?  
A) phosphorus  
B) potassium  
C) argon  
D) aluminum
- What is the total number of occupied principal energy levels in a neutral atom of neon in the ground state?  
A) 1  
B) 2  
C) 3  
D) 4

- Which sublevel configuration correctly represents a completely filled third principal energy level?  
A)  $3s^2 3p^5 3d^0$   
B)  $3s^2 3p^2 3d^{10}$   
C)  $3s^2 3p^6 3d^{10}$   
D)  $3s^2 3p^6 3d^6$
- Which atom in the ground state contains a partially filled 3p orbital?  
A) argon  
B) calcium  
C) potassium  
D) aluminum
- In the ground state, the atoms of elements in Period 2 all have the same number of  
A) protons  
B) neutrons  
C) 1s electrons  
D) oxidation states
- Which is the electron configuration for a neutral atom in the ground state?  
A)  $1s^2 2s^2 3s^1$   
B)  $1s^2 2s^2 2p^2 3s^1$   
C)  $1s^2 2s^2 2p^6 3p^1$   
D)  $1s^2 2s^2 2p^6 3s^1$
- Which represents the electron configuration of an isotope of oxygen in the ground state?  
A)  $1s^2 2s^2 2p^1$   
B)  $1s^2 2s^2 2p^2$   
C)  $1s^2 2s^2 2p^2$   
D)  $1s^2 2s^2 2p^6$
- The total number of sublevels in the fourth principal energy level of an atom is  
A) 1  
B) 2  
C) 3  
D) 4
- Which is a possible electron configuration for argon in the excited state?  
A)  $1s^2 2s^2 2p^7 3s^2 3p^5$   
B)  $1s^2 2s^2 2p^5 3s^2 3p^6$   
C)  $1s^2 2s^2 2p^6 3s^1 3p^7$   
D)  $1s^2 2s^2 2p^6 3s^2 3p^5 4s^1$
- Which is a correct description of the shape and spatial orientation of the p orbitals in an atom?  
A) All have the same shape but a different orientation.  
B) All have the same shape and the same orientation.  
C) All have a different shape and a different orientation.  
D) All have a different shape but the same orientation.
- A maximum of 6 electrons can occupy  
A) an s orbital  
B) an s sublevel  
C) a p orbital  
D) a p sublevel
- How does the ground state electron configuration of the hydrogen atom differ from that of a ground state helium atom?  
A) Hydrogen has one electron in a higher energy level.  
B) Hydrogen has two electrons in a lower energy level.  
C) Hydrogen contains a half-filled orbital.  
D) Hydrogen contains a completely filled orbital.

# Electron Configuration Review Worksheet Answers

**Christian G. Meyer**



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Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall  
Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction     **Strategies for Information Technology Governance** Wim Van Grembergen,2004-01-01 The advent of the Information Society is marked by the explosive penetration of information technologies in all aspects of life and by a related fundamental transformation in every form of the organization Researchers business people and policy makers have recognized the importance of addressing technological economic and social impacts in conjunction For example the rise and fall of the dot com hype depended a lot on the strength of the business model on the technological capabilities available to firms and on the readiness of the society and economy at large sustain a new breed of business activity However it is notoriously difficult to examine the cross impacts of social economic and technological aspects of the Information Society This kind of work requires multidisciplinary work and collaboration on a wide range of skills Social and Economic Transformation in the Digital Era addresses this challenge by assembling the latest thinking of leading researchers and policy makers The book covers all key subject areas of the Information Society an presents innovative business models case studies normative theories and social explanations     **Mission-Critical Security Planner** Eric Greenberg,2003-01-30 Shows step by step how to complete a customized security improvement plan including analyzing needs justifying budgets and selecting technology while dramatically reducing time and cost Includes worksheets at every stage for creating a comprehensive security plan meaningful to management and technical staff Uses practical risk management techniques to intelligently assess and manage the network security risks facing your organization Presents the material in a witty and lively style backed up by solid business planning methods Companion Web site provides all worksheets and the security planning template     Merrill Chemistry Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998     CISSP Training Guide Roberta Bragg,2003 The CISSP Certified Information Systems Security Professionals exam is a six hour monitored paper based exam covering 10 domains of information system security knowledge each representing a specific area of expertise This book maps the exam objectives and offers numerous features such as exam tips case studies and practice exams     *InfoWorld* ,1982-04-19 InfoWorld is targeted to Senior IT professionals Content is segmented into Channels and Topic Centers InfoWorld also celebrates people companies and projects     **Target IBPS Bank PO/ MT Preliminary & Main Exams 20 Practice Sets Workbook - 16 in Book + 4 Online (6th edition)** Disha Experts,2017-07-05 Target IBPS Bank Preliminary Main PO MT Exam 20 Practice Sets Workbook with 4 Online Tests is the

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