

Model 1 - White Light

	•	Color	Photon Energy (x 10 <sup>-21</sup> ) (J)	Wavelength Range (nm)	Speed (m/s)
Lightbulb (white light)	Prism	Reds	269-318	625-740	3.00 × 10°
		Oranges	318-337	590-625	3.00 × 10 <sup>8</sup>
	34 97	Yellows	337-352	565-590	3.00 × 10 <sup>4</sup>
( *	$\wedge$ /	Greens	352-382	520-565	$3.00 \times 10^{8}$
10/	$\checkmark \setminus \mathscr{U}$	Blues	382-452	440-520	$3.00 \times 10^{6}$
		Violets	452-523	380-440	$3.00 \times 10^{4}$

- 1. Trace the arrows in Model 1 and shade in the table with colored pencils where appropriate.
- 2. What happens to white light when it passes through a prism?

3. Why are the color labels in the table in Model 1 plural (i.e., "Reds" rather than "Red")?

4. Do all colors of light travel at the same speed?

5. Do all colors of light have the same energy? If no, which colors have the highest energy and the least energy, respectively?

warm colors have less energy cool colors have more

# **Electron Energy And Light Answer Key**

Hitendra K. Malik

#### **Electron Energy And Light Answer Key:**

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