

# 8 • Electron Configurations & Periodicity

## WRITING ELECTRON CONFIGURATIONS

For each given element, fill in the orbital diagram and then write the electron configuration for the element.

1.	2.	3.	4.	5.	6.
Element: Ar # of e <sup>-</sup> 's: <u>18</u>	Element: Mg # of e <sup>-</sup> 's: <u>12</u>	Element: N # of e <sup>-</sup> 's: <u>7</u>	Element: Li # of e <sup>-</sup> 's: <u>3</u>	Element: P # of e <sup>-</sup> 's: <u>15</u>	Element: Cl # of e <sup>-</sup> 's: <u>17</u>

Write the electron configurations of each of these in **long form** and **short form**:

- Ar 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup> 3s<sup>2</sup> 3p<sup>6</sup>  
 Ar [Ne] 3s<sup>2</sup> 3p<sup>6</sup>
- Mg 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup> 3s<sup>2</sup>  
 Mg [Ne] 3s<sup>2</sup>
- N 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>3</sup>  
 N [He] 2s<sup>2</sup> 2p<sup>3</sup>
- Li 1s<sup>2</sup> 2s<sup>1</sup>  
 Li [He] 2s<sup>1</sup>
- P 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup> 3s<sup>2</sup> 3p<sup>3</sup>  
 P [Ne] 3s<sup>2</sup> 3p<sup>3</sup>
- Cl 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup> 3s<sup>2</sup> 3p<sup>5</sup>  
 Cl [Ne] 3s<sup>2</sup> 3p<sup>5</sup>



7. Fill in the orbital diagram for the element, Fe, and write the electron configuration of Fe in the long and short form.



# Electron Configuration And Periodicity Lab Answer Key

**JE Gale**



**Electron Configuration And Periodicity Lab Answer Key:**

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