

Electron Configuration Lab

Name _____ KEY _____ Period _____ Date _____

Directions: To understand that electrons fill up the LOWEST energy orbitals first, but that they spread out as much as possible in orbitals.

Part 1: Questions

1. Write the Aufbau Principle.

Electrons will fill the lowest energy orbitals first.

2. Which "rule" or "principle" states that electrons spread out as much as possible within a sublevel before they pair up in the same orbital? **Hund's Rule**

3. In this lab, one cup represents an orbital in the atom. How many cups would be needed to represent the "2p" sublevel?

three (3)

4. How many cups would be needed to represent an atom that has the 1s, 2s, 2p, 3s, 3p sublevels? _____?

ten (10)

5. In this lab, beans represent the electrons. How many beans can go into a single cup?

two (2)

Part 2:

Part 2a

Electron Configuration	Total # of Electrons	Element Name	Teacher's Initials
$1s^2 2s^2 2p^6 3s^2 3p^3$	15	Phosphorus	

Part 2b

Station	Electron Configuration	Total # of Electrons
Station 1	$1s^2 2s^2 2p^6$	10
Station 2	$1s^2 2s^2 2p^6$	8
Station 3	$1s^2 2s^2 2p^6 3s^2 3p^2 4s^2$	20
Station 4	$1s^2 2s^2 2p^6 3s^2 3p^6$	18
Station 5	$1s^2 2s^2 2p^3$	7
Station 6	$1s^2 2s^2$	4
Station 7	$1s^2 2s^2 2p^3$	6
Station 8	$1s^2 2s^2$	4
Station 9	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2$	18
Station 10	$1s^2 2s^2 2p^3$	7

Part 2c: Questions

1. Using a Periodic Table, identify the elements at each station and complete the table to the right.

2. Which orbital is lower in energy -- 4s or 3d?

4s

3. Write the electron configuration for Zinc.

$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10}$

4. Write the electron configuration for Xenon.

$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6$

5. Write the electron configuration for Bismuth.

$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^3$

Station	Element Name
Station 1	Neon (Ne)
Station 2	Oxygen (O)
Station 3	Calcium (Ca)
Station 4	Argon (Ar)
Station 5	Boron (B)
Station 6	Beryllium (Be)
Station 7	Carbon (C)
Station 8	Lithium (Li)
Station 9	Potassium (K)
Station 10	Nitrogen (N)

Electron Configuration Lab

Christian G. Meyer

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the name bar.

Electron Configuration Lab:

This book delves into Electron Configuration Lab. Electron Configuration Lab is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Electron Configuration Lab, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Electron Configuration Lab
 - Chapter 2: Essential Elements of Electron Configuration Lab
 - Chapter 3: Electron Configuration Lab in Everyday Life
 - Chapter 4: Electron Configuration Lab in Specific Contexts
 - Chapter 5: Conclusion
2. In chapter 1, this book will provide an overview of Electron Configuration Lab. The first chapter will explore what Electron Configuration Lab is, why Electron Configuration Lab is vital, and how to effectively learn about Electron Configuration Lab.
3. In chapter 2, the author will delve into the foundational concepts of Electron Configuration Lab. The second chapter will elucidate the essential principles that must be understood to grasp Electron Configuration Lab in its entirety.
4. In chapter 3, this book will examine the practical applications of Electron Configuration Lab in daily life. The third chapter will showcase real-world examples of how Electron Configuration Lab can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Electron Configuration Lab in specific contexts. This chapter will explore how Electron Configuration Lab is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Electron Configuration Lab. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Electron Configuration Lab.

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