# DISCLAIMER: THIS IS A SAMPLE OF QUESTIONS/PROBLEMS FROM THIS YEAR. PLEASE BE SURE TO REVIEW YOUR NOTES, LABS, COJ'S, OBJECTIVE SHEETS, ETC. <u>DO NOT RELY ONLY ON THIS REVIEW GUIDE.</u>

		Thoice choice that best completes the statement of	и англиен	s the question.	
	1)	Which of the following observations is qua. The liquid turns blue litmus paper red b. The liquid boils at 100°C. c. The liquid tastes bitter. d. The liquid is cloudy.			
	2)	The symbols for units of length in order from largest to smallest are			
		a. m, cm, mm, km.	C.	km, mm, cm, m.	
		b. mm, m, cm, km.	d.	km, m, cm, mm.	
	3)	A volume of 1 milliliter is equivalent to			
		<ul> <li>a. I cubic centimeter.</li> </ul>	C.	1 liter.	
		b. I gram.	d.	10 <sup>-1</sup> cubic decimeters.	
	40	Which pair of quantities determines the d	a material?		
		a. mass and weight	C.	volume and concentration	
		b. volume and weight	d.	volume and mass	
	5)	A sample of gold has a mass of 96.5 g an	ne of 5.00 cm3. The density of gold is		
		<ul> <li>a. 0.0518 g/cm<sup>3</sup>.</li> </ul>	C.	101.5 g/cm <sup>3</sup> .	
		<ul> <li>b. 19.3 g/cm³.</li> </ul>	d.	483 g/cm <sup>3</sup> .	
	6)	The number of grams equal to 0.5 kg is			
		a. 0.0005.	C.	500.	
		b. 0.005.	d.	5000.	
	7)	If 1 inch equals 2.54 cm, how many centi	pual 1 yard?		
		a. 0.0706 cm	e.	30.5 cm	
		b. 14.2 cm	d.	91.4 cm	
_	8)	the true mass of the sample, it can be said that these values have			
		<ul> <li>a. good precision.</li> </ul>			
		<ul> <li>b. good accuracy.</li> </ul>			
		c. poor precision.			
		d. poor accuracy.			
_	9)				
		a. 0.03 g.		0.036 g.	
		b. 0.35 g.	d.	$3.5 \times 10^{2}$ g.	
	1.00	When 64.4 is divided by 2.00, the correct	mumber c	of significant figures in the result is	

# **Lab Chem Study Guide 2012 Final Exam**

**L** Cohen

Lab Chem Study Guide 2012 Final Exam:

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In a digitally-driven world wherever displays reign great and quick conversation drowns out the subtleties of language, the profound techniques and psychological nuances concealed within phrases frequently get unheard. Yet, situated within the pages of **Lab Chem Study Guide 2012 Final Exam** a charming fictional prize sporting with raw emotions, lies an exceptional journey waiting to be undertaken. Penned by a talented wordsmith, this charming opus encourages viewers on an introspective journey, softly unraveling the veiled truths and profound influence resonating within the fabric of every word. Within the mental depths of this moving evaluation, we shall embark upon a sincere exploration of the book is core subjects, dissect their fascinating publishing fashion, and yield to the effective resonance it evokes strong within the recesses of readers hearts.

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