

Experiment 6B: Calculations with a Chemical Reaction

Objectives:

1. To understand the theory of stoichiometry, including the concept of limiting reagents and percent yield, for homogeneous catalyst reactions.
2. To calculate the limiting reagent, including the limiting reagent present in the solution.
3. To calculate stoichiometry for catalyst that is not used.
4. To calculate the theoretical maximum amount of product that could be produced.
5. To calculate the percent yield of the actual amount of product, including the limiting reagent.

Materials: Please refer to Experiment 6B page 95 of the Pharmacokinetics Laboratory.

Equipment:

Procedure:

1. Prepare very carefully experiment with the proper stoichiometry.
2. Prepare a catalyst (10% solution) and a product (10% solution).
3. Calculate 10 ml of catalyst (10% solution) and 10 ml of product (10% solution).
4. Mix the catalyst and the product (10 ml) and then the product (10 ml) and the catalyst (10 ml).
5. Use the catalyst and the product (10 ml) and the product (10 ml).
6. Prepare 10 ml of catalyst (10% solution) and 10 ml of product (10% solution).
7. Use the catalyst and the product (10 ml) and the product (10 ml).
8. Use the catalyst and the product (10 ml) and the product (10 ml).
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