

## UNIT III - THERMOCHEMISTRY

1. Calculate the H-F bond energy given the following data.



$$\Delta H_{\text{rxn}} = -544 \text{ kJ}$$

- A. 1130 kJ/mol    B. 565 kJ/mol    **C. 283 J/mol**    D. 300 kJ/mol    E. 134 kJ/mol
2. 1.0 g of the rocket fuel hydrazine ( $\text{N}_2\text{H}_4$ ) is burned in a bomb calorimeter. The temperature of the bomb calorimeter changes from 24.63 °C to 28.16 °C. The heat capacity of the bomb calorimeter is 837 J/K. Calculate the heat of combustion per mol of hydrazine in the bomb calorimeter.

- A. 95.0 kJ/mol**    B. 2.96 kJ/mol    C.  $3.55 \times 10^3$   
D. 92.4 J/mol    E.  $1.13 \times 10^5$  kJ/mol

3. Calculate the enthalpy change for the following reaction of nitric oxide NO with ozone
- $\text{O}_3$
- .



Given the following data:



- A. -1083.7 kJ    B. -693.8 kJ    C. -198.8 kJ    **D. -93.7 kJ**    E. -93.7 kJ
4. An endothermic reaction does no work on the surroundings. What values do you predict for the heat and energy change of this reaction?

- A.  $q < 0$ ,  $\Delta E > 0$     **B.  $q < 0$ ,  $\Delta E < 0$**     C.  $q > 0$ ,  $\Delta E > 0$   
D.  $q > 0$ ,  $\Delta E < 0$     E.  $q > 0$ ,  $\Delta E = 0$
5. M&M candies contain 209.0 Calories per 1.5 oz serving. The fat content accounts for 38.3% of these Calories. What is the energy of the fat content in kJ?

- A. 19.1 kJ    B. 874 kJ    **C.  $1.91 \times 10^{-2}$  kJ**  
D.  $3.35 \times 10^5$  kJ    E. 335 kJ
6. Calculate the heat released when 2.00 g of butane from a cigarette lighter is burned.



- A. -90.7 kJ    **B. -98.3 kJ**    C. -49.2 kJ    D. -108.4 kJ    E. -196.6 kJ
7. When Karl Kaverman adds chilled grog to a granite mug, 10.9 kJ of energy is transferred from the mug to the grog. If the mug has a mass of 625 g and an initial temperature of 25 °C, what is the final temperature of the mug? The specific heat of granite is 0.79 J/g·K.

- A. 2.9 °C    B. 14 °C    **C. 22 °C**    D. 25 °C    E. 47 °C

# General Chemistry Multiple Choice Questions And Answers

**RD Boyd**



## **General Chemistry Multiple Choice Questions And Answers:**

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## The Enigmatic Realm of **General Chemistry Multiple Choice Questions And Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **General Chemistry Multiple Choice Questions And Answers** a literary masterpiece penned by a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those that partake in its reading experience.

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