

CHEM EXAM 3 Fall 2009 - Form A

Fill in your name, CID and section on this test booklet. Fill in and bubble in your name, CID (for the letter "C" bubble zero), section and "001" on the scantron form. For questions #66, mark A. For each question, read all possible choices then mark the best one on your scantron. Use a #2 pencil and make clean answers to insure proper scoring. When finished, turn in your scantron and show this booklet and your CID to the proctor. All questions are worth 4 points each. Let your knowledge show!

1. Which of the following is balanced correctly?

- A. $\text{CH}_4 + 2 \text{O}_2 \rightarrow \text{CO}_2 + 2 \text{H}_2\text{O}$
 B. $6 \text{H}_2\text{O} + \text{P}_4\text{O}_{10} \rightarrow 4 \text{H}_3\text{PO}_4$
 C. $3 \text{H}_2 + 3 \text{N}_2 \rightarrow 2 \text{NH}_3$
 D. $2 \text{C}_2\text{H}_6\text{O} + 3 \text{O}_2 \rightarrow 2 \text{CO}_2 + 4 \text{H}_2\text{O}$

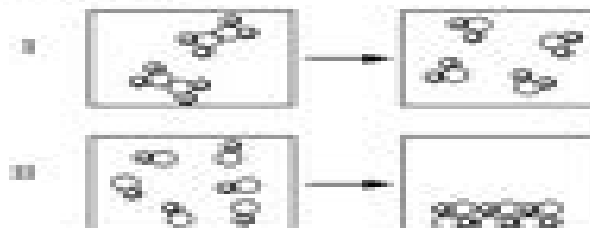
A chemical equation is balanced when the same number of atoms are in the reactants and products. The reactants in B contain 16 oxygen atoms, but the products contain only 12 (B). The reactants in C contain 6 nitrogen atoms, but the product contains only 2 (C). The reactants in D contain 6 carbon atoms, 16 hydrogen atoms, and 8 oxygen atoms. The products contain 2 carbon atoms, 8 oxygen atoms, and 8 hydrogen atoms (D).

2. Which of the following is an oxidation-reduction reaction?

- A. $2 \text{NaCl (aq)} + \text{Ca(NO}_3)_2 \text{ (aq)} \rightarrow 2 \text{NaNO}_3 \text{ (aq)} + \text{CaCl}_2 \text{ (aq)}$
 B. $\text{CrCl}_3 \text{ (aq)} + 3 \text{KOH (aq)} \rightarrow 3 \text{KCl (aq)} + \text{Cr(OH)}_3 \text{ (s)}$
 C. $\text{KOH (aq)} + \text{HCl (aq)} \rightarrow \text{KCl (aq)} + \text{H}_2\text{O (l)}$
 D. $2 \text{CrCl}_3 \text{ (aq)} + \text{CuCl}_2 \text{ (aq)} \rightarrow 2 \text{CrCl}_2 \text{ (aq)} + \text{Cu (s)}$

An oxidation-reduction reaction is one in which electrons are transferred from one or more atoms in the reactants to one or more atoms in the products. The reactants and the products in A are all strong electrolytes (a net ionic equation, A). Two aqueous solutions react in B to form a solid precipitate, B. A strong acid reacts with a strong base in C to form an ionic salt and water (acid-base, C). Two aqueous solutions react to form a solid in D, but chromium in CrCl_3 has an oxidation number of +3. In CrCl_2 , it has an oxidation number of +2. Copper in CuCl_2 has an oxidation number of +2. Solid copper has an oxidation number of 0. Therefore, chromium is oxidized while copper is reduced (an oxidation-reduction reaction).

3. Which of the following transfers heat to the surroundings?



- A. I only
 B. II only
 C. Both I and II
 D. neither I or II

In Figure I, the bond between the two white spheres is broken. In Figure II, six molecules combine to form a solid precipitate. Bond breaking requires energy (endothermic), while bond forming releases energy (exothermic).

4. How many grams of N_2H_4 would be required to react completely with 19.0 g of O_2 according to the balanced chemical equation?



- A. 37.1 g
 B. 42.1 g
 C. 74.2 g
 D. 23.2 g
 E. 26.5 g

The chemical equation is used to relate the number of moles of one substance to the number of moles of another. If grams of O_2 is given, it needs to be converted to moles. The coefficients are used as mole ratios, and the formula mass of N_2H_4 is used to get back to grams:

$$\begin{aligned} 19.0 \text{ g O}_2 & \left(\frac{1 \text{ mol O}_2}{32.0 \text{ g O}_2} \right) \left(\frac{2 \text{ mol N}_2\text{H}_4}{3 \text{ mol O}_2} \right) \left(\frac{32.0 \text{ g N}_2\text{H}_4}{1 \text{ mol N}_2\text{H}_4} \right) \\ & = 37.3 \text{ g N}_2\text{H}_4 \end{aligned}$$

General Chemistry Lab Answers

Thomas Elert



General Chemistry Lab Answers:

Exploring General Chemistry in the Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science *Laboratory Manual for Principles of General Chemistry* Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

Course Success in the Undergraduate General Chemistry Lab Thomas Elert, 2019-11-15 Stetig hohe Studienabbruchquoten in den MINT F chern an deutschen Hochschulen welche auch aus geringem Kurserfolg in einf hrenden Laborpraktika resultieren k nnten und die wachsende Kritik an der Qualit t und Wirksamkeit ebendieser machen eine eingehende Betrachtung von Laborpraktika notwendig Diese Studie untersuchte die Lernziele des Laborpraktikums Allgemeine Chemie f r Lehramtsstudierende im ersten Semester sowie Faktoren f r den Kurserfolg um daraus Aussagen ber den Stellenwert von Laborpraktika in der universit ren Bildung insbesondere f r langfristigen Studienerfolg abzuleiten Dazu wurde ein theoretisches Modell zu Grunde gelegt welches das Vorwissen der Studierenden und die Lernzielpassung zwischen Studierenden und Lehrenden als zwei entscheidende Faktoren f r Kurserfolg ber cksichtigt

Constantly high student dropout rates in STEM subjects at German universities which could be the result of low course success in introductory laboratory courses among other things and increasing criticism about their quality and effectiveness necessitate these laboratory courses to be examined thoroughly This study investigated the learning goals of the General Chemistry laboratory course for first year students in teacher training and factors for course success in order to make statements about the significance of laboratory courses for university education particularly for long term study success For this purpose a theoretical model that assumes the students prior knowledge and learning goal alignment between students and their lab instructors to be two defining factors for lab course success was used as a framework

The United States Catalog Mary Burnham, Carol Hurd, 1928 **Teaching Undergraduate Science** Linda C. Hodges, 2023-07-03 This book is written for all science or engineering faculty who have ever found themselves baffled and frustrated by their undergraduate students lack of engagement and learning The author an experienced scientist faculty member and educational consultant addresses these issues with the knowledge of faculty interests constraints and day to day

concerns in mind Drawing from the research on learning she offers faculty new ways to think about the struggles their science students face She then provides a range of evidence based teaching strategies that can make the time faculty spend in the classroom more productive and satisfying Linda Hodges reviews the various learning problems endemic to teaching science explains why they are so common and persistent and presents a digest of key ideas and strategies to address them based on the research she has undertaken into the literature on the cognitive sciences and education Recognizing that faculty have different views about teaching different comfort levels with alternative teaching approaches and are often pressed for time Linda Hodges takes these constraints into account by first offering a framework for thinking purposefully about course design and teaching choices and then providing a range of strategies to address very specific teaching barriers whether it be students motivation engagement in class ability to problem solve their reading comprehension or laboratory research or writing skills Except for the first and last chapters the other chapters in this book stand on their own i e can be read in any order and address a specific challenge students have in learning and doing science Each chapter summarizes the research explaining why students struggle and concludes by offering several teaching options categorized by how easy or difficult they are to implement Some for example can work in a large lecture class without a great expenditure of time others may require more preparation and a more adventurous approach to teaching Each strategy is accompanied by a table categorizing its likely impact how much time it will take in class or out and how difficult it will be to implement Like scientific research teaching works best when faculty start with a goal in mind plan an approach building on the literature use well tested methodologies and analyze results for future trials Linda Hodges message is that with such intentional thought and a bit of effort faculty can succeed in helping many more students gain exciting new skills and abilities whether those students are potential scientists or physicians or entrepreneurs Her book serves as a mini compendium of current research as well as a protocol manual a readily accessible guide to the literature the best practices known to date and a framework for thinking about teaching

America's Lab Report National Research Council, Division of Behavioral and Social Sciences and Education, Center for Education, Board on Science Education, Committee on High School Laboratories: Role and Vision, 2006-01-20 Laboratory experiences as a part of most U S high school science curricula have been taken for granted for decades but they have rarely been carefully examined What do they contribute to science learning What can they contribute to science learning What is the current status of labs in our nation s high schools as a context for learning science This book looks at a range of questions about how laboratory experiences fit into U S high schools What is effective laboratory teaching What does research tell us about learning in high school science labs How should student learning in laboratory experiences be assessed Do all student have access to laboratory experiences What changes need to be made to improve laboratory experiences for high school students How can school organization contribute to effective laboratory teaching With increased attention to the U S education system and student outcomes no part of the high school curriculum

should escape scrutiny This timely book investigates factors that influence a high school laboratory experience looking closely at what currently takes place and what the goals of those experiences are and should be Science educators school administrators policy makers and parents will all benefit from a better understanding of the need for laboratory experiences to be an integral part of the science curriculum and how that can be accomplished *WAC and Second Language Writers* Terry Myers Zawacki,Michelle Cox,2014-05-14 Editors and contributors pursue the ambitious goal of including within WAC theory research and practice the differing perspectives educational experiences and voices of second language writers The chapters within this collection not only report new research but also share a wealth of pedagogical curricular and programmatic practices relevant to second language writers Representing a range of institutional perspectives including those of students and faculty at public universities community colleges liberal arts colleges and English language schools and a diverse set of geographical and cultural contexts the editors and contributors report on work taking place in the United States Asia Europe and the Middle East Journal of Chemical Education ,1926 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers **Projects in Higher Education** National Science Foundation (U.S.),1976 Instructional Technology Theory in the Post-Pandemic Era Carbonara, David D.,Tomei, Lawrence A.,2024-07-22 The COVID 19 Pandemic transformed nearly every aspect of daily life across the globe in just a few short years Thankfully we ve made it a long way from the days of no contact social distancing masks and general isolation Still many aspects of this time have continued into the present This is particularly true regarding education which saw a massive overhaul during that period Remote learning and technology infused education were a necessity then and may prove to be an invaluable improvement as we go forward Instructional Technology Theory in the Post Pandemic Era investigates the facets of incorporating technology and virtual spaces into education permanently The experienced educators that compiled this book utilize their years of knowledge to bring to light the intricacies of adapting virtual education laboratories for the foreseeable future They examine student performance metrics detail teacher development practices consider the social aspects of tech infused education and explore the implementation of new pedagogies for best results Covering topics such as companionship in distance education pandemic teaching experiences and professional and teacher development this book is a valuable resource for educators pre service teachers administrators policymakers academicians researchers and more **General Catalogue** Berea College,1907 **Book of Abstracts** American Chemical Society. Meeting,American Chemical Society,1987 **Teaching Innovation in University Education: Case Studies and Main Practices** Saura, Jose Ramon,2022-06-17 In the last decade the development of new technologies has made innovation a fundamental pillar of education Teaching innovation includes the evolution of both teaching and learning models to drive improvements in educational methodologies Teaching innovation is a pioneer in the understanding and comprehension of the different teaching methodologies and models developed in the academic area Teaching innovation

is a process that seeks validation in the academic and teaching communities at universities in order to promote the improvement and its practices and uses in the future characterized by digital development and data based methods Teaching Innovation in University Education Case Studies and Main Practices features the major practices and case studies of teaching innovation developed in recent years at universities It is a source on study cases focused on teaching innovation methodologies as well as on the identification of new technologies that will help the development of initiatives and practices focused on teaching innovation at higher education institutions Covering topics such as didactic strategics service learning and technology based gamification this premier reference source is an indispensable resource for pre service teachers lecturers students faculty administrators libraries entrepreneurs researchers and academicians *Modern Hospital* ,1924

Biennial Catalog New Mexico Institute of Mining and Technology,New Mexico. Institute of Mining and Technology,1893

Homeschooling High School Jeanne Gowen Dennis,2004 Homeschooling High School 2nd Edition Planning Ahead for College Admission gives you the tools you need to design a solid high school program that both maximizes educational freedom and leads to successful college admission Find answers to your questions about grading scales diplomas entrance exams scholarships application procedures and much more all while maintaining your family s ideal balance between autonomy and accountability in your home high school

HCI in Games: Serious and Immersive Games Xiaowen Fang,2021-07-03 This two volume set LNCS 12789 and 12790 constitutes the refereed proceedings of the Third International Conference on HCI in Games HCI Games 2021 held as part of the 23rd International Conference HCI International 2021 which took place in July 2021 Due to COVID 19 pandemic the conference was held virtually The total of 1276 papers and 241 posters included in the 39 HCII 2021 proceedings volumes was carefully reviewed and selected from 5222 submissions The papers of HCI Games 2021 Part II are organized in topical sections named Serious Games Gamification and Learning Mixed and Virtual Reality Games

Bulletin of the University Summer Session University of Florida,1929 **Annual Report** Massachusetts Agricultural College,1913 **College Science Improvement Programs; COSIP A & B Report** National Science Foundation (U.S.). Office of Experimental Programs,1974

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