

DIRECTIONS: Type your answers on answer sheet located on last page.
Due No Later Than 11:00 AM Wednesday, Feb. 3rd.

1. Atoms of the same element, copper for example, have the same number of _____.
A. electrons in the nucleus
B. neutrons in the outer nuclear shell
C. protons in the nucleus
D. electrons in the outer energy level
2. Which of the following correctly describes isotopes of the same element?
A. The atoms have different numbers of protons and the same number of neutrons
B. The atoms have the same number of electrons and different numbers of protons
C. The atoms have different numbers of neutrons and the same number of protons
3. Which one of the following is *not* true for minerals?
A. They form naturally.
B. They can be a liquid, solid, or gas.
C. They have a specific, predictable chemical composition.
D. They are inorganic.
4. Based on the definition of a mineral, which one of the following is *not* considered to be a mineral?
A. halite (table salt) B. quartz C. talc D. sugar
5. What of these is the hardest known mineral in nature?
A. silicate B. gold C. diamond D. mica
6. The resistance of a mineral to abrasion (being scratched) is known as _____.
A. luster B. cleavage C. streak D. hardness
7. Which group of minerals are the most abundant in the Earth's crust?
A. sulfides B. carbonates C. silicates D. halides
8. The tendency of some minerals to break along smooth, parallel planes is known as _____.
A. streak B. cleavage C. luster D. crystal habit
9. The most variable physical property of minerals such as quartz is _____.
A. hardness B. crystal habit C. specific gravity D. color
10. The most abundant element by weight in the Earth's crust is _____.
A. aluminum B. iron C. oxygen D. hydrogen

11. The most abundant mineral in the Earth's crust is mica.
12. Luster is a physical property that describes the light reflecting characteristics of a mineral.
13. A mineral can be composed entirely of one element.
14. Streak is the physical property that shows the color of a mineral in its powdered form.
15. Rocks, like minerals, are always composed of inorganic matter.

Environmental Science 20practice Exam Answers

**Andrew Friedland, Rick Relyea, David
Courard-Hauri**

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