

NAME: _____ BLOCK: _____



Elements & Macromolecules in Organisms

Most common elements in living things are **carbon, hydrogen, nitrogen, and oxygen**. These four elements constitute about **95% of your body weight**. All compounds can be classified in two broad categories --- **organic and inorganic compounds**. Organic compounds are made primarily of **carbon**. Carbon has **four outer electrons** and can form four bonds. Carbon can form **single bonds** with another atom and also bond to other carbon molecules forming **double, triple, or quadruple bonds**. Organic compounds also contain **hydrogen**. Since hydrogen has only one electron, it can form only **single bonds**.

Each small organic molecule can be a unit of a large organic molecule called a **macromolecule**. There are **four classes of macromolecules** (polysaccharides or **carbohydrates**, triglycerides or **lipids**, polypeptides or **proteins**, and nucleic acids such as DNA & RNA). **Carbohydrates and lipids** are made of only carbon, hydrogen, and oxygen (**CHO**). **Proteins** are made of carbon, hydrogen, oxygen, and nitrogen (**CHON**). **Nucleic acids** such as DNA and RNA contain carbon, hydrogen, oxygen, nitrogen, and phosphorus (**CHON P**).

Use the drawing of the **amino acid on this worksheet** (look ahead to another page for this sketch and remember that the **NUMBER OF LINES** from a single atom is their **NUMBER OF BONDS**) to determine the number of bonds formed by:

_____ Oxygen _____ Hydrogen _____ Nitrogen

The body also needs trace amounts of other elements such as calcium, potassium, and sulfur for proper functioning of muscles, nerves, etc.

Questions:

1. Name the 4 main elements that make up 95% of an organism.
2. Name the 4 types of bonds carbon can form.
3. What are macromolecules?
4. Name the 4 classes of macromolecules.
5. Give 2 examples of nucleic acids. _____
6. What elements make up carbohydrates & lipids (symbols)? _____
7. Name 3 elements your body needs trace amounts of for proper functioning.

Elements Macromolecules In Organisms Packet Answers

**Dave Wessner, Christine
Dupont, Trevor Charles, Josh Neufeld**

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SPBE, Self-pacing Biology Experiences James L. Kelly, Alan R. Orr, 1980 *Microbiology* Dave Wessner, Christine Dupont, Trevor Charles, Josh Neufeld, 2017-08-28 Microbiology 2nd Edition helps to develop a meaningful connection with the material through the incorporation of primary literature applications and examples The text offers an ideal balance between comprehensive in depth coverage of core concepts while employing a narrative style that incorporates many relevant applications and a unique focus on current research and experimentation The book frames information around the three pillars of physiology ecology and genetics which highlights their interconnectedness and helps students see a bigger picture This innovative organization establishes a firm foundation for later work and provides a perspective on real world applications of microbiology Bulletin of the Atomic Scientists , 1961-05 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin s iconic Doomsday Clock stimulates solutions for a safer world **Kos** , 1993 *Elements in Living Organisms* Suzanne Slade, 2006-07-15 Explore the four elements carbon nitrogen oxygen and hydrogen that are the major building blocks of life on Earth *Atoms, Molecules & Elements: What Are Molecules? Gr. 5-8* George Graybill, 2015-10-01 This is the chapter slice What Are Molecules from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives **Water and Biological Macromolecules** Eric Westhof, 1993

Periodic Tables Unifying Living Organisms at the Molecular Level A. Lima-de-Faria, 2017 The DNA sequencing of a series of living organisms has elucidated many biological problems But the internal atomic and electronic evolution of DNA remains to be mapped in detail RNA and DNA now appear to be the prime determinants of biological evolution leading to the sudden appearance of novel organism structures and functions that emerge ready made as a surprise to the organism This has been demonstrated by the manipulation of genes that led to the sudden production of additional complete wings and legs in flies and birds The study of this internal atomic construction of macromolecules is being investigated at the large electron accelerators such as the MAX IV Synchrotron Radiation Laboratory Lund University Sweden The periodicity of the chemical elements is well known from its iconic Table Significantly this periodicity can now be seen to extend to the properties of

living organisms Biological properties as different as flight vision luminescence and regeneration as well as others show unexpectedly periodic emergence They resurface without previous announcement in most unrelated plant and animal families and they emerge irrespective of whether the organism is a simple invertebrate or a most complex mammal Moreover this periodicity does not necessarily start at the cell or DNA levels but appears initially in crystals and minerals where it is shown to be a pure atomic and electronic process e g in luminescence and regeneration The assembled molecular evidence led to the construction of Periodic Tables of living organisms placing them in a position comparable to the periodicity of the chemical elements Surprisingly there are striking resemblances between the periodicities of the chemical elements and those of living organisms In addition the two types of Tables increase our insight into the events directing atomic evolution since the periodic law established in chemical elements turns out to be applicable to the periodicity of living organisms The new Periodic Tables introduce a predictive capacity in biological evolution that before was hardly contemplated Eric Scerri from the Department of Chemistry and Biochemistry California University Los Angeles who is the Author of the book The Periodic Table Its Story and its Significance Oxford University Press stated in an e mail that Professor Lima de Faria s book is wonderful and a pioneering work **Atoms, Molecules & Elements Gr. 5-8** George Graybill,2007-09-01 Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource makes the periodic table easier to understand Begin by answering what are atoms See how the atomic model is made up of electrons protons and neutrons Find out what a molecule is and how they differ from elements Then move on to compounds Find the elements that make up different compounds Get comfortable with the periodic table by recognizing each element as part of a group Examine how patterns in the period table dictate how those elements react with others Finally explore the three important kinds of elements metals nonmetals and inert gases Aligned to the Next Generation Science Standards and written to Bloom s Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included **Atoms, Molecules & Elements: Properties of Important Elements Gr. 5-8** George Graybill,2015-10-01 This is the chapter slice Properties of Important Elements from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives **Atoms, Molecules & Elements: What Are**

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Macromolecules Containing Metal and Metal-like Elements Alaa S. Abd-El-Aziz,2003

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