

ANSWER KEY

ATOMS AND ELECTRONS STUDY GUIDE

Learning Goals: (1) (10-11 P. 41) Explain a new scientific theory of atoms (the modern atomic theory) by describing changes in the atomic model over time, including those changes based on experimental observations. (10-11 P. 42) Explain the scientific theory of atomic orbitals (model for atomic theory) by describing the scientific changes in terms of quantum mechanical calculations, including wave-particle duality and how it relates to some of their uses, including charges and locations within the atom. (10-11 P. 42) Explain properties of atoms and their placement within periodic table based on properties of their electrons, related to Pauli exclusion principle, Hund's rule, Aufbau principle (building-up principle), spectroscopy, mass spectrometry, and others, using models.

ANSWER KEY

Directions: Match the atomic theory selected on the correct card(s).

Model/Model	Scientist(s)	Atomic Theory's Basic Model(s)
1C	Democritus (460 BC)	A. Everything is made of atoms that cannot be created, divided or destroyed
4A	Bohr (1913)	B. When Electrons Absorb energy, usually in packets, charged matter is accelerated and travels throughout the positive material
1B	Thomson (1897)	C. "atomic indivisible (uncuttable)"
7	Rutherford (1911)	D. Group Number (Valence Electrons)
1D	Wheeler (1927)	E. Electrons Cloud
1E	Schrodinger (1926)	F. Atoms randomly occupy space the subatomic structure model

Directions: Use the periodic table to answer the following questions:

- 10-11** = **10-11** The number of protons equals the number of electrons.
- atomic mass minus the atomic number will equal the **Number of neutrons**.
- 10-11** = **10-11** protons are positive (positively charged) electrons are negative (negatively charged).

Electrons In Atoms Guided Practice Problems Answers

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Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step by step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid base reactions, which often persists through working with acid base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields. **The Practice of Chemistry Study Guide & Solutions Manual** Pamela

Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes. Also includes solutions and explanations to odd numbered exercises. A Visual Analogy Guide to Chemistry, 2e Paul A

Krieger, 2018-02-01 A Visual Analogy Guide to Chemistry is the latest in the innovative and widely used series of books by Paul Krieger. This study guide delivers a big picture view of difficult concepts and effective study tools to help students learn and understand the details of general organic and biochemistry topics. A Visual Analogy Guide to Chemistry is a worthwhile investment for any introductory chemistry student. A Level Chemistry MCQ (Multiple Choice Questions) Arshad

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