

1. Which electron configuration represents an atom in an excited state?

- A)  $1s^2 2s^2 2p^6 3p^1$                       C)  $1s^2 2s^2 2p^6 3s^2 3p^2$   
 B)  $1s^2 2s^2 2p^6 3s^2 3p^1$                       D)  $1s^2 2s^2 2p^6 3s^2$

2. All of the elements in Period 3 have a total of 2 electrons in the

- A) 2s sublevel                      C) 2p sublevel  
 B) 3s sublevel                      D) 3p sublevel

3. Which atom in the ground state has three unpaired electrons in its outermost principal energy level?

- A) Li                                      C) N  
 B) B                                      D) Ne

4. What is the total number of valence electrons in an atom with the electron configuration  $1s^2 2s^2 2p^6 3s^2 3p^3$ ?

- A) 6                                      C) 3  
 B) 2                                      D) 5

5. Which orbital notation represents an atom of beryllium in the ground state?

A) 

<b>1s</b>	<b>2s</b>	<b>2p</b>		
↑	↑	↑	↑	

B) 

↑↓	↑	↑		
----	---	---	--	--

C) 

↑↓	↑↓			
----	----	--	--	--

D) 

↑		↑	↑	↑
---	--	---	---	---

6. Which element has a completely filled third principal energy level?

- A) Ar                                      C) Fe  
 B) N                                      D) Zn

7. Which atom in the ground state has only 3 electrons in the 3p sublevel?

- A) phosphorus                      C) argon  
 B) potassium                      D) aluminum

8. What is the total number of occupied principal energy levels in a neutral atom of neon in the ground state?

- A) 1                                      C) 3  
 B) 2                                      D) 4

9. Which sublevel configuration correctly represents a completely filled third principal energy level?

- A)  $3s^2 3p^6 3d^6$                       C)  $3s^2 3p^6 3d^{10}$   
 B)  $3s^2 3p^2 3d^{10}$                       D)  $3s^2 3p^6 3d^6$

10. Which atom in the ground state contains a partially filled 3p orbital?

- A) argon                                  C) potassium  
 B) calcium                              D) aluminum

11. In the ground state, the atoms of elements in Period 2 all have the same number of

- A) protons                              C) 1s electrons  
 B) neutrons                              D) oxidation states

12. Which is the electron configuration for a neutral atom in the ground state?

- A)  $1s^2 2s^2 3s^1$                       C)  $1s^2 2s^2 2p^6 3p^1$   
 B)  $1s^2 2s^2 2p^2 3s^1$                       D)  $1s^2 2s^2 2p^6 3s^1$

13. Which represents the electron configuration of an isotope of oxygen in the ground state?

- A)  $1s^2 2s^2 2p^1$                       C)  $1s^2 2s^2 2p^2$   
 B)  $1s^2 2s^2 2p^2$                       D)  $1s^2 2s^2 2p^4$

14. The total number of sublevels in the fourth principal energy level of an atom is

- A) 1                                      C) 3  
 B) 2                                      D) 4

15. Which is a possible electron configuration for argon in the excited state?

- A)  $1s^2 2s^2 2p^7 3s^2 3p^2$                       C)  $1s^2 2s^2 2p^6 3s^1 3p^2$   
 B)  $1s^2 2s^2 2p^2 3s^2 3p^6$                       D)  $1s^2 2s^2 2p^6 3s^2 3p^5 4s^1$

16. Which is a correct description of the shape and spatial orientation of the p orbitals in an atom?

- A) All have the same shape but a different orientation.  
 B) All have the same shape and the same orientation.  
 C) All have a different shape and a different orientation.  
 D) All have a different shape but the same orientation.

17. A maximum of 6 electrons can occupy

- A) an s orbital                      C) a p orbital  
 B) an s sublevel                      D) a p sublevel

18. How does the ground state electron configuration of the hydrogen atom differ from that of a ground state helium atom?

- A) Hydrogen has one electron in a higher energy level.  
 B) Hydrogen has two electrons in a lower energy level.  
 C) Hydrogen contains a half-filled orbital.  
 D) Hydrogen contains a completely filled orbital.

# Electron Distribution Review Worksheet Answers

**J Rink**



**Electron Distribution Review Worksheet Answers:**

The Top Books of the Year Electron Distribution Review Worksheet Answers The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous captivating novels enthraling the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the fascinating narratives that have enthralled audiences this year. Electron Distribution Review Worksheet Answers : Colleen Hoover's "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. Electron Distribution Review Worksheet Answers : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Electron Distribution Review Worksheet Answers : Delia Owens' "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These bestselling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a masterful and gripping novel that will keep you wondering until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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