

1. For **each** of the following sets of sublevels, determine which has the lowest energy, and which has the highest energy.

(a) 2s, 3s, 4s

(b) 4f, 4p, 4d

(c) 4d, 4f, 3d

(d) 6s, 3d, 5f

Ex for (a) - Lowest 2s, highest 4s

2. Explain what  $6s^2$  means.

3. What is the maximum number of electrons that can be found in an s – sublevel? \_\_\_\_ p – sublevel? \_\_\_\_ d – sublevel? \_\_\_\_ f – sublevel? \_\_\_\_

4. Indicate which of these (there may be more than one) orbital designations is incorrect:

(a) 4s

(b) 3f

(c) 2d

(d) 3d

5. Draw energy level diagrams for

(a) sodium

(b) silicon

(c) beryllium

(d) cobalt

6. How many unpaired electrons are there in each of the atoms in question 5?

7. Give the electron configuration for:

(a) lithium

(b) oxygen

(c) selenium

(d) iodine

(e) silver

(f) arsenic

(g) indium

(h) potassium

8. Give the name and symbol for the atom whose electron configuration corresponds to each of the following electron configurations.

a)  $1s^2 2s^2 2p^6 3s^2 3p^5$

b)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^3$

c)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^6$

d)  $1s^2 2s^2 2p^6 3s^2 3p^2$

e)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$

9. Write the shorthand notation for the above electron configurations.

# Electron Configuration Extension Questions

**Gary Wulfsberg**



## Electron Configuration Extension Questions:

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*Theoretical Chemistry* R N Dixon, C Thomson, 2007-10-31 Specialist Periodical Reports provide systematic and detailed review coverage of progress in the major areas of chemical research Written by experts in their specialist fields the series creates a unique service for the active research chemist supplying regular critical in depth accounts of progress in particular areas of chemistry For over 90 years The Royal Society of Chemistry and its predecessor the Chemical Society have been publishing reports charting developments in chemistry which originally took the form of Annual Reports However by 1967 the whole spectrum of chemistry could no longer be contained within one volume and the series Specialist Periodical Reports was born The Annual Reports themselves still existed but were divided into two and subsequently three volumes covering Inorganic Organic and Physical Chemistry For more general coverage of the highlights in chemistry they remain a must Since that time the SPR series has altered according to the fluctuating degree of activity in various fields of chemistry Some titles have remained unchanged while others have altered their emphasis along with their titles some have been combined under a new name whereas others have had to be discontinued The current list of Specialist Periodical Reports can be seen on the inside flap of this volume

### **Coupling of Biological and Electronic Systems**

Karl-Heinz Hoffmann, 2002-07-09 With the F caesarium that was held in Bonn in November of 1999 the newly founded research center caesar center of advanced european studies and research has initiated a series of symposia which are to bring researchers together for discussions and definition of positions on the various research fields at caesar on an annual basis The areas of research at caesar incorporate materials science nanotechnology coupling of biological and electronic systems and ergonomics in communications and In November of 2000 the 2 caesarium this time with the theme of Coupling of Biological and Electronic Systems was held in Bonn The conference transcript at hand delivers an impression of the introduced topics and discussions The coupling of biochemics and electronics together with bioinformatics is an exciting and current thematic complex which will surely co define the development of science and research for a long time The presentations held at the 2 caesarium gave a glimpse into broadly diversified areas such as bioinformatics biomolecules and biosensor technology Further miniaturisation up to the nanoscale played as much of a role as the conception of fast algorithms for the structural reconnoitring of proteins We would like to thank the participants of the symposium for the inspiring discussions in a not always easy interdisciplinary dialogue Thanks to the organising committee the colleagues

Michael Famulok Daniel Hoffmann Michael Moske and Eckhard Quandt the symposium became a success **Chemistry**  
**insights 'O' level** Rex M. Heyworth,2007 **Advances in Quantum Chemistry** ,1964-01-01 Advances in Quantum  
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