

Molarity Calculations

Calculate the molarities of the following solutions:



- 1) 2.3 moles of sodium chloride in 0.45 liters of solution.

$$M = \frac{2.3 \text{ moles}}{0.45 \text{ L}} = 5.111 \frac{\text{mol}}{\text{L}} = \boxed{5.11 \text{ M}}$$



- 2) 1.2 moles of calcium carbonate in 1.22 liters of solution.

$$M = \frac{1.2 \text{ mol}}{1.22 \text{ L}} = 0.983606557 \frac{\text{mol}}{\text{L}} = \boxed{0.98 \text{ M}}$$



- 3) 0.09 moles of sodium sulfate in 12 mL of solution.

12 mL = 0.012 L

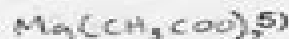
$$M = \frac{0.09 \text{ mol}}{0.012 \text{ L}} = \boxed{7.5 \text{ M}}$$



- 4) 0.75 moles of lithium fluoride in 65 mL of solution.

65 mL = 0.065 L

$$M = \frac{0.75 \text{ mol}}{0.065 \text{ L}} = 11.53846154 \text{ M} = \boxed{11.54 \text{ M}}$$



- 5) 0.8 moles of magnesium acetate in 5 liters of solution.

$$M = \frac{0.8 \text{ mol}}{5 \text{ L}} = \boxed{0.16 \text{ M}}$$



- 6) 120 grams of calcium nitrite in 240 mL of solution.

240 mL = 0.240 L

$$n = 120 \text{ g} \left( \frac{1 \text{ mol}}{132.09 \text{ g}} \right)$$

MW<sub>Ca(NO<sub>2</sub>)<sub>2</sub></sub> = 132.09 g/mol

n = 0.91 mol

$$M = \frac{0.91 \text{ mol}}{0.240 \text{ L}} = \boxed{3.79 \text{ M}}$$



- 7) 98 grams of sodium hydroxide in 2.2 liters of solution.

MW<sub>NaOH</sub> = 39.997 g/mol

$$n = 98 \text{ g} \left( \frac{1 \text{ mol}}{39.997 \text{ g}} \right) = 2.45 \text{ mol}$$

$$M = \frac{2.45 \text{ mol}}{2.2 \text{ L}} = \boxed{1.11 \text{ M}}$$



- 8) 1.2 grams of hydrochloric acid in 25 mL of solution.

MW<sub>HCl</sub> = 36.461 g/mol

$$n = (1.2 \text{ g}) \left( \frac{1 \text{ mol}}{36.461 \text{ g}} \right) = 0.033 \text{ mol}$$

$$M = \frac{0.033 \text{ mol}}{0.025 \text{ L}} = \boxed{1.32 \text{ M}}$$



- 9) 45 grams of ammonia in 0.75 L of solution.

MW<sub>NH<sub>3</sub></sub> = 17.031 g/mol

$$n = 45 \text{ g} \left( \frac{1 \text{ mol}}{17.031 \text{ g}} \right) = 2.64 \text{ mol}$$

$$M = \frac{2.64 \text{ mol}}{0.75 \text{ L}} = \boxed{3.52 \text{ M}}$$

# Molarity Calculations Key Answers Show Work

**Leo J. Malone, Theodore Dolter**



## **Molarity Calculations Key Answers Show Work:**

*Chemistry: 1,001 Practice Problems For Dummies (+ Free Online Practice)* Heather Hattori, Richard H.

Langley, 2014-04-14 Practice makes perfect and helps deepen your understanding of chemistry Every high school requires a course in chemistry and many universities require the course for majors in medicine engineering biology and various other sciences 1001 Chemistry Practice Problems For Dummies provides students of this popular course the chance to practice what they learn in class deepening their understanding of the material and allowing for supplemental explanation of difficult topics 1001 Chemistry Practice Problems For Dummies takes you beyond the instruction and guidance offered in Chemistry For Dummies giving you 1 001 opportunities to practice solving problems from the major topics in chemistry Plus an online component provides you with a collection of chemistry problems presented in multiple choice format to further help you test your skills as you go Gives you a chance to practice and reinforce the skills you learn in chemistry class Helps you refine your understanding of chemistry Practice problems with answer explanations that detail every step of every problem Whether you re studying chemistry at the high school college or graduate level the practice problems in 1001 Chemistry Practice Problems For Dummies range in areas of difficulty and style providing you with the practice help you need to score high at exam time *E3 Chemistry Guided Study Book - 2018 Home Edition (Answer Key Included)* Effiong Eyo, 2017-12-08

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Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance     **Introductory Chemistry**

Charles H. Corwin, 2005 For one semester courses in Basic Chemistry Introduction to Chemistry and Preparatory Chemistry and the first term of Allied Health Chemistry This text is carefully crafted to help students learn chemical skills and concepts more effectively Corwin covers math and problem solving early in the text he builds student confidence and skills through innovative problem solving pedagogy and technology formulated to meet student needs     Methods in Molecular Biophysics

Nathan R. Zaccai, Igor N. Serdyuk, Joseph Zaccai, 2017-05-18 Current techniques for studying biological macromolecules and their interactions are based on the application of physical methods ranging from classical thermodynamics to more recently developed techniques for the detection and manipulation of single molecules Reflecting the advances made in biophysics research over the past decade and now including a new section on medical imaging this new edition describes the physical methods used in modern biology All key techniques are covered including mass spectrometry hydrodynamics microscopy and imaging diffraction and spectroscopy electron microscopy molecular dynamics simulations and nuclear magnetic resonance Each method is explained in detail using examples of real world applications Short asides are provided throughout to ensure that explanations are accessible to life scientists physicists and those with medical backgrounds The book remains an unparalleled and comprehensive resource for graduate students of biophysics and medical

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**Analytical Chemistry** Bryan M. Ham, Aihui MaHam, 2015-10-01 A comprehensive study of analytical chemistry providing the basics of analytical chemistry and introductions to the laboratory Covers the basics of a chemistry lab including lab safety glassware and common instrumentation Covers fundamentals of analytical techniques such as wet chemistry instrumental analyses spectroscopy chromatography FTIR NMR XRF XRD HPLC GC MS Capillary Electrophoresis and proteomics Includes ChemTech an interactive program that contains lesson exercises useful calculators and an interactive periodic table Details Laboratory Information Management System a program used to log in samples input data search samples approve samples and print reports and certificates of analysis

**Clinical Laboratory Science - E-Book** Mary Louise Turgeon, 2022-09-14 Selected for Doody's Core Titles 2024 in Laboratory Technology Using a discipline by discipline approach Turgeon's Clinical Laboratory Science Concepts Procedures and Clinical Applications 9th Edition provides a fundamental overview of the concepts procedures and clinical applications essential for working in a clinical laboratory and performing routine clinical lab tests Coverage includes basic laboratory techniques and key topics such as safety phlebotomy quality assessment automation and point of care testing as well as discussion of clinical laboratory specialties Clear straightforward instructions simplify laboratory procedures and are guided by the latest practices and CLSI Clinical and Laboratory Standards Institute standards Written by well known CLS educator Mary Louise Turgeon this edition offers essential guidance and recommendations for today's laboratory testing methods and clinical applications Broad scope of coverage makes this text an ideal companion for

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The Practice of Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink [http://www.bfwpub.com/pdfs/wink/POCPowerPoint\\_Final.ppt](http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt) 832KB

**Clinical Chemistry** Michael L. Bishop, Edward P. Fody, Larry E. Schoeff, 2013-02-20 In its Seventh Edition this acclaimed Clinical Chemistry continues to be the most student friendly clinical chemistry text available This edition not only covers the how of clinical testing but also places greater emphasis on the what why and when in order to help today's students fully understand the implications of the information covered as well as the applicability of this crucial topic in practice With clear explanations that strike just the right balance of analytic principles techniques and correlation of results with disease states this edition has been fully updated with the latest information to help keep today's students at the forefront of today's science New case studies practice questions and exercises provide ample opportunities to review and apply the topics covered through the text



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