



HP
Procurve Switches
2524, 2512, and 4108gl
Command Line Interface
Reference Guide

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July 2001

Hp Pro Curve 2524 Manual

O García



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 $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$
 $2 \text{ NaCl} + \text{F}_2 \rightarrow 2 \text{ NaF} + \text{Cl}_2$
 $2 \text{ H}_2\text{O} + 5 \text{ Pb} \rightarrow 2 \text{ H}_2 + 5 \text{ PbO}$
 $2 \text{ HCl} + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{Cl}^-$
 $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 4 \text{ KBr} + \text{Al}_2\text{SO}_4 + 3 \text{ K}_2\text{SO}_4$
 $\text{CH}_4 + 2 \text{ O}_2 \rightarrow \text{CO}_2 + 2 \text{ H}_2\text{O}$
 $\text{C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$

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 $1 \text{ O}_2 + 2 \text{ NH}_3 \rightarrow \text{H}_2\text{O} + 2 \text{ O}_2 + 3 \text{ H}_2$
 $2 \text{ O}_2 + 2 \text{ H}_2\text{O} \rightarrow 4 \text{ RbH} + 2 \text{ O} + \text{RbOH}$
 $2 \text{ H}_2 + 5 \text{ Cl}_2 \rightarrow 2 \text{ KBr} + \text{KCl} + \text{Br}_2$
 $6 \text{ N}_2 + 2 \text{ H}_2 + \text{NH}_3 \rightarrow 7 \text{ N}_2\text{F}_6 + 3 \text{ Ca} + \text{N}_2\text{F}_6 + 2 \text{ H}_2 + 5 \text{ OH}$

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 $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$
 $2 \text{ NaCl} + \text{F}_2 \rightarrow 2 \text{ NaF} + \text{Cl}_2$
 $2 \text{ H}_2\text{O} + 5 \text{ Pb} \rightarrow 2 \text{ H}_2 + 5 \text{ PbO}$
 $2 \text{ HCl} + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{Cl}^-$
 $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 4 \text{ KBr} + \text{Al}_2\text{SO}_4 + 3 \text{ K}_2\text{SO}_4$
 $\text{CH}_4 + 2 \text{ O}_2 \rightarrow \text{CO}_2 + 2 \text{ H}_2\text{O}$

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 $4 \text{ Fe} + 3 \text{ O}_2 \rightarrow 2 \text{ Fe}_2\text{O}_3$
 $2 \text{ Fe}_2\text{O}_3 + 3 \text{ H}_2\text{SO}_4 \rightarrow 1 \text{ Fe}_2(\text{SO}_4)_3 + 3 \text{ H}_2\text{O}$
 $\text{C}_4\text{H}_{10} + 6 \text{ O}_2 \rightarrow 4 \text{ CO}_2 + 5 \text{ H}_2\text{O}$

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 $3 \text{ Fe} + 2 \text{ O}_2 \rightarrow 3 \text{ FeO}$
 $2 \text{ Fe} + 3 \text{ CO} \rightarrow 2 \text{ Fe}_3\text{O}_4 + 3 \text{ H}_2$
 $3 \text{ BaO} + 3 \text{ PO}_4 \rightarrow 2 \text{ H}_2 + 4 \text{ O} + 2 \text{ MgNH}_4\text{PO}_4$
 $4 \text{ Mg} + 2 \text{ P} \rightarrow 2 \text{ O}_7 + 2 \text{ NH}_3$
 $3 \text{ H}_2 + 2 \text{ O} + 4 \text{ Al} \rightarrow 2 \text{ SO}_4 + 2 \text{ Al}_2\text{SO}_4 + 3 \text{ H}_2$
 $4 \text{ P} + 10 \text{ HCl} \rightarrow 3 \text{ H}_3\text{PO}_4 + 3 \text{ H}_2 + 4 \text{ NO}_2$
 $2 \text{ O}_3 + 2 \text{ H}_2\text{NO} \rightarrow 2 \text{ H}_2\text{O} + 2 \text{ HNO}_2$

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