In 2009, Thomas Edison invented the nickel-iron battery. The following: reaction takes place in the battery.

Fe +
$$2810(00H) + 2H(0) \rightarrow Fe(0H)q + $288(0H)q$$$

What is the limiting reactant when 5.00 mol of Fe and 6.00 mol of NEOCOSHS: trisact?

How many make of Fe(COI) a will be produced?

How many moles of the other reaction will be in except?

g) One of the few sense compounds that form is certain somes, heptafluoride. OCSTRONET YOU.

What is the limiting reactant in the reaction of sax mol of cerium fluoride with so,o mol of senon beautypridelt

How many an less of CoXeF year, be produced?

How many make of the other reactant will be expens?

so I run is obtained commercially by the reaction of hematite (Fe pO to with carbon monoraide.

What is the limiting reactant when properly of hematite reacts with poor mod of carbon monopolis?

How management of Fe can be produced?

Limiting Reactants And Percent Yield 9 3 Answer Key

Allan Blackman, Daniel Southam, Gwendolyn Lawrie, Natalie Williamson, Christopher Thompson

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