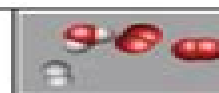


Activity A:**Water reaction**Get the Gizmo ready:

- Make sure $\text{H}_2 + \text{O}_2$ becomes H_2O is selected.
- Click **Reset** (↺).

**Goal: Identify a limiting reactant**

1. Count: H_2O is the **chemical formula** for water. In order to produce a single molecule of water, how many hydrogen atoms are needed? Oxygen atoms?

2. Predict: Set the number of O_2 molecules to five and the number of H_2 molecules to eight.

- A. How many oxygen atoms are present? Hydrogen atoms?
- B. How many water molecules could form from these reactants?
- C. After the molecules react, which reactant will be left over?
- D. Which reactant will be the limiting reactant?
- E. Click **Play** and wait until **Reaction complete** is shown. What happened?

Due to the hydrogen and oxygen reacting, there are eight water molecules and one remaining oxygen molecule.

3. Test: Click **Reset**. Set the number of O_2 molecules to 10.

- A. How many hydrogen molecules (H_2) will be needed for there to be no excess reactants? Use the slider to set the number of H_2 molecules to this value.

- B. How many water molecules will be formed?

- C. Click **Play**. What happened?

4. Challenge yourself: Run the Gizmo with five O_2 molecules and nine H_2 molecules.

- A. How many of each molecule do you think will be produced by this reaction?

O_2 :

H_2 :

H_2O :

- B. Click **Play**. How many were actually produced? O_2 :

H_2 :

H_2O :

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J Rink

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'J Rink' text.

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