

Naming Compounds Practice Problems Naming Compounds: Select from the following list: Acid, Base, Binary Compound, Ionic Compound, Molecular Compound, Organic Compound, or Other.

Name the following compounds:

- 1) HNO_3 : **nitric acid**
- 2) NaOH : **sodium hydroxide**
- 3) H_2SO_4 : **sulfuric acid**
- 4) NaCl : **sodium chloride**
- 5) H_2O : **water**
- 6) H_2SO_3 : **sulfurous acid**
- 7) HCl : **hydrochloric acid**
- 8) Na_2SO_4 : **sodium sulfate**
- 9) NaOH : **sodium hydroxide**
- 10) H_2SO_4 : **sulfuric acid**
- 11) NaCl : **sodium chloride**
- 12) H_2SO_3 : **sulfurous acid**
- 13) HNO_3 : **nitric acid**
- 14) NaOH : **sodium hydroxide**

Write the formulas for the following compounds:

- 1) sodium chloride: **NaCl**
- 2) ammonium sulfate: **$(\text{NH}_4)_2\text{SO}_4$**
- 3) nitric acid: **HNO_3**
- 4) ammonium hydroxide: **NH_4OH**
- 5) silver chloride: **AgCl**
- 6) sodium nitrate: **NaNO_3**
- 7) ammonium chloride: **NH_4Cl**
- 8) calcium chloride: **CaCl_2**
- 9) sodium hydroxide: **NaOH**
- 10) ammonium nitrate: **NH_4NO_3**

Hs Science Unit Lesson Answers

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Decoding **Hs Science Unit Lesson Answeres**: Revealing the Captivating Potential of Verbal Expression

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