Part A

Calculate the amount of heat (in kilojoules) required to vaporize 2.58 kg of water at its boiling point.

Express the heat in kilojoules to three significant figures.

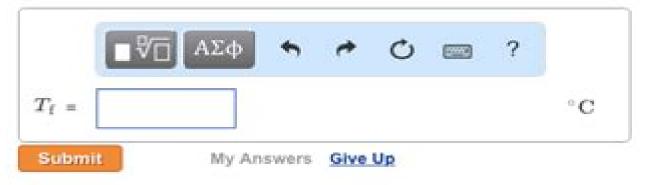
Correct

At its boiling point, the enthalpy of vaporization ($\Delta H_{\rm vap}$) of water is 40.7 kJ/mol. Once the given mass, 2.58×10^3 g, is converted to the number of moles using the molar mass of water (18.02 g/mol), multiplying the number of moles by $\Delta H_{\rm vap}$ yields the amount of heat required to evaporate that amount of water.

Part B

Suppose that 0.48 g of water at 25° C condenses on the surface of a 55-g block of aluminum that is initially at 25° C. If the heat released during condensation goes only toward heating the metal, what is the final temperature (in degrees Celsius) of the metal block? (The specific heat capacity of aluminum, $C_{s,A1}$, is 0.903 $J/(g \cdot {}^{\circ}C)$.)

Express the temperature in degrees Celsius to two significant figures.



Mastering Chemistry Answers Chapter 12

John S. Phillips, Cheryl Wistrom

Mastering Chemistry Answers Chapter 12:

CBSE Class 12 Mastering Organic Reactions MCQ With Answers Priti Singhal, 2024-11-11 The primary objective of this book is to serve as a comprehensive guide for students educators and researchers by focusing on reaction mechanisms practical applications and problem solving techniques Organic chemistry is not just about memorizing equations and formulas it is about understanding how molecules interact change and influence each other under different conditions With that in mind this book emphasizes the logic and patterns behind organic reactions making it easier for readers to apply concepts across a variety of scenarios Each chapter of this book builds upon foundational knowledge ensuring a progressive learning experience From nucleophilic substitutions to pericyclic reactions and from oxidation reduction mechanisms to named reactions we cover both fundamental and advanced topics to cater to students at all levels Real world examples have been integrated throughout the chapters to show how organic reactions play essential roles in pharmaceuticals biochemistry agriculture and environmental science This approach bridges the gap between theory and practical applications helping readers appreciate the relevance of organic chemistry in daily life **Chemistry** John Olmsted, Gregory M. Williams, 1997 Textbook outling concepts of molecular science General Chemistry for Engineers Jeffrey Gaffney, Nancy Marley, 2017-11-13 General Chemistry for Engineers explores the key areas of chemistry needed for engineers This book develops material from the basics to more advanced areas in a systematic fashion As the material is presented case studies relevant to engineering are included that demonstrate the strong link between chemistry and the various areas of engineering Serves as a unique chemistry reference source for professional engineers Provides the chemistry principles required by various engineering disciplines Begins with an atoms first approach building from the simple to the more complex chemical concepts Includes engineering case studies connecting chemical principles to solving actual engineering problems Links chemistry to contemporary issues related to the interface between chemistry and engineering practices

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