

16. The no. of Neutrons in  $^{14}\text{N}$  d) 9  
 a) 6 b) 7 c) 8
17. The energy of one quantum of radiation is calculated by d)  $E = h\nu$   
 a)  $E = h\nu$  b)  $h = E\nu$  c)  $h = \nu/E$

(08)

### Q.2) Answer the followings.

- What are Physical classification of Matter?
- Define the elements and explains?
- Define Atomic Mass?
- Write the chemical equation for combustion of propane?
- Write down the IUPAC names?  
 $\text{CH}_3\text{--C}(\text{C}_2\text{H}_5)_2\text{--CH}_3$   $\text{C}(\text{CH}_3)_4\text{--CH}_3$
- Define the law of conservation of mass?
- If an element X has mass no. 11 and it has 6 neutrons then write its representation.
- State Heisenberg Uncertainty Principle?

### Section- B

(16)

#### Attempt any Eight

- Q.3) How will you convert Propane into @-nitro propane?
- Q.4) How will you convert ethane into ethyl chloride?
- Q.5) State the uses of alkane?
- Q.6) write the physical properties of alkanes?
- Q.7) What are alkane? how they are classified?
- Q.8) Find the no. of Moles (n) present in 72 grams of  $\text{H}_2\text{O}$ ?
- Q.9) Write down the Units of Measurements?
- Q.10) Differentiates between Isotopes and Isobars?
- Q.11) State and Explains Pauli's exclusion principle
- Q.12) State Hund's rule of maximum Multiplicity?
- Q.13) State the order of fillings Atomic Orbital's model

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