

Chapter Test B, continued

10. In a reaction, the loss of two unpaired electrons placed in separate orbitals to form two new unpaired electrons. These reactions are called a
- spin-forbidden reaction.
 - spin-allowed low reaction.
 - spin-forbidden reaction.
 - double-spin-forbidden reaction.
11. The use of a double arrow in a chemical equation indicates that the reaction
- is reversible.
 - requires heat.
 - is highly exothermic.
 - has not been confirmed in the laboratory.

PART II Write the correct term (or terms) in the space provided.

12. A critical value is reached that requires a change of a formula in a chemical equation is called (a) _____.
13. In a chemical equation, the symbol (l) indicates that the substance is _____.
14. The product in a formula for a liquid reactant is (a) _____.
15. The general equation for a synthesis reaction is _____.
16. The general equation for a decomposition reaction is _____.
17. The general equation for a single-displacement reaction is _____.
18. The general equation for a double-displacement reaction is _____.
- _____
- _____

Holt Science Spectrum The Nature Of Chemical Reactions Answers

SJ Ball



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