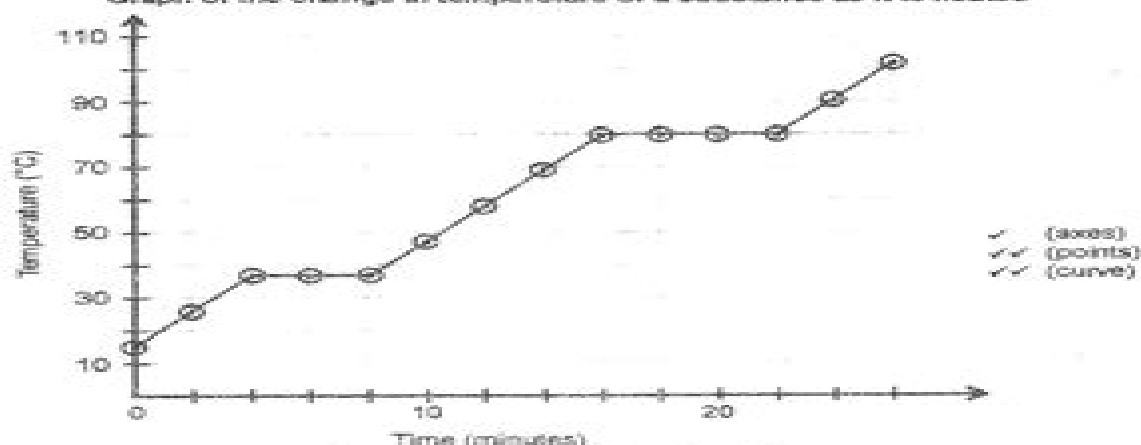


3.4

Graph of the change in temperature of a substance as it is heated

(5)



- 3.5 Substance exists as both a solid and a liquid ✓✓ or... is melting (2)  
 3.6 37 °C ✓ (2)  
 3.7 80 °C ✓ (1)  
 3.8 An indication of the average kinetic energy ✓ of the particles in a substance. ✓ (2)  
 3.9 Physical ✓ (1)  
 3.10 In the solid state the intermolecular forces are the strongest and the particles cannot move. The heater provides thermal energy to the system, which is translated into kinetic energy (increase in temperature) until the intermolecular forces between the particles can be broken. ✓✓ At this point the thermal energy is transformed into latent energy to break the bonds, so no increase in temperature occurs. ✓✓ (4)

[23]

## QUESTION 4

- 4.1  
 4.1.1 Homogeneous ✓ (1)  
 4.1.2 Homogeneous ✓ (1)  
 4.1.3 Pure substance ✓ (1)  
 4.2  
 4.2.1 Potassium bromide ✓ (2)  
 4.2.2 Calcium carbonate ✓ (2)  
 4.2.3 Ammonium chloride ✓ (2)  
 4.3  
 4.3.1  $\text{ZnSO}_4$  ✓✓ (2)  
 4.3.2  $\text{KMnO}_4$  ✓✓ (2)  
 4.3.3  $\text{H}_2\text{O}$  ✓✓ (2)

[17]

# Grade March 2014 Life Sciences Question Paper

**AW Rasmussen**



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## **Table of Contents Grade March 2014 Life Sciences Question Paper**

1. Understanding the eBook Grade March 2014 Life Sciences Question Paper
  - The Rise of Digital Reading Grade March 2014 Life Sciences Question Paper
  - Advantages of eBooks Over Traditional Books
2. Identifying Grade March 2014 Life Sciences Question Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Grade March 2014 Life Sciences Question Paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from Grade March 2014 Life Sciences Question Paper
  - Personalized Recommendations
  - Grade March 2014 Life Sciences Question Paper User Reviews and Ratings
  - Grade March 2014 Life Sciences Question Paper and Bestseller Lists
5. Accessing Grade March 2014 Life Sciences Question Paper Free and Paid eBooks
  - Grade March 2014 Life Sciences Question Paper Public Domain eBooks
  - Grade March 2014 Life Sciences Question Paper eBook Subscription Services
  - Grade March 2014 Life Sciences Question Paper Budget-Friendly Options
6. Navigating Grade March 2014 Life Sciences Question Paper eBook Formats
  - ePub, PDF, MOBI, and More
  - Grade March 2014 Life Sciences Question Paper Compatibility with Devices
  - Grade March 2014 Life Sciences Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Grade March 2014 Life Sciences Question Paper
  - Highlighting and Note-Taking Grade March 2014 Life Sciences Question Paper
  - Interactive Elements Grade March 2014 Life Sciences Question Paper

8. Staying Engaged with Grade March 2014 Life Sciences Question Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Grade March 2014 Life Sciences Question Paper
9. Balancing eBooks and Physical Books Grade March 2014 Life Sciences Question Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Grade March 2014 Life Sciences Question Paper
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Grade March 2014 Life Sciences Question Paper
  - Setting Reading Goals Grade March 2014 Life Sciences Question Paper
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Grade March 2014 Life Sciences Question Paper
  - Fact-Checking eBook Content of Grade March 2014 Life Sciences Question Paper
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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