

# NAMING COMPOUNDS: PRACTICE

Name the following covalent compounds:

SO<sub>3</sub>      sulfur trioxide

N<sub>2</sub>S      dinitrogen <sup>mono</sup>sulfide

P<sub>2</sub>Br<sub>4</sub>      diphosphorus tetrabromide

CO      carbon monoxide (not ~~monoxide~~)

NO<sub>2</sub>      nitrogen dioxide

# Formula Writing For Covalent Compounds

**Thirumalaisamy P.**  
**Velavancorresponding**



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