

Equation Writing and Predicting Products

Introduction:

If you examine a car that has been in a junkyard for a while, you will notice that it has rusted. Rusting is a slow chemical reaction of the iron in the car's body with oxygen gas. If a piece of sodium is put into water, a much more rapid chemical reaction occurs. Sodium reacts with water to produce sodium hydroxide and hydrogen gas. During this reaction, enough heat is liberated to ignite the hydrogen gas, causing it to explode.

Chemists observe what is happening in a chemical reaction and try to describe it in language that is simple and clear. A chemical equation uses formulas and symbols to describe the substances involved in a reaction, the physical state of the substance, the use of a catalyst, and relative proportions. The general form of an equation is:



In this investigation you will perform a series of reactions and make careful observations of the changes that occur. Using simple tests and your knowledge of chemistry, you will determine the identity of the products. With this information, you will write chemical equations to describe the reactions.

Pre-Lab Discussion:

Read through the entire lab investigation and the relevant pages of your textbook. Then answer the questions that follow.

- What constitutes a positive test for each of the following gases?
 - oxygen _____
 - hydrogen _____
 - carbon dioxide _____
 - water vapor _____
 - ammonia (NH₃) _____
- What is the proper way to smell a substance in the lab? Why should care be taken when smelling a gas such as ammonia? _____
- What is the role of a catalyst in a reaction? How can you tell when a substance serves as a catalyst? (This answer will be important later, spend some time on it now.) _____
- One way to identify limestone (CaCO₃) is to drop a small amount of hydrochloric acid (HCl) on it. A positive test results in a fire of carbon dioxide being produced. Write out in words the information represented by the following balanced chemical equation.



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S Ben Porath



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General, Organic, and Biochemistry Lab Manual Sara Selfe, 2006-01-12 Offers a choice of classic chemistry experiments and innovative ones All of them place special emphasis on the biological implications of chemical concepts Available for custom publishing at <http://custompub.whfreeman.com> **Lab Manual for General, Organic, and Biochemistry** Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives **Essentials of General Organic and Biochemistry** captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit www.whfreeman.com/gob

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management and land reclamation A standard text in a variety of courses the Techniques Manual as it is commonly called covers every aspect of modern wildlife management and provides practical information for applying the hundreds of methods described in its pages This deft and thorough update ensures that The Wildlife Techniques Manual will remain an indispensable resource one that professionals and students in wildlife biology conservation and management simply cannot do without *Merrill Chemistry-Lab.Manual* Smoot,1994-07 **Addison-Wesley Small-scale Chemistry** Dennis D.

Staley,Edward L. Waterman,1995 **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science** ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Modern Packaging* ,1963

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Lehman,2009 This comprehensive lab companion provides enough theory to help students understand how and why an operation works but emphasizes the practical aspects of an operation to help them perform the operation successfully in the lab The new 4th edition contains a new introductory section Chemistry and the Environment which includes a discussion of the principles of green chemistry Several green experiments have been added and some experiments from the previous editions have been revised to make them greener *Scientific and Technical Aerospace Reports* ,1994 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database *Popular Mechanics* ,2000-01 Popular Mechanics inspires instructs and influences readers to help them master the modern world Whether it s practical DIY home improvement tips gadgets and digital technology information on the newest cars or the latest breakthroughs in science PM is the ultimate guide to our high tech lifestyle

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