

You may want to advance (1) [exam 5.05](#), (4) [section 14.2](#) while completing this module.

A 0.10 M aqueous solution of an ionic compound with the formula MX has a freezing point of -2.3°C .

Calculate the van't Hoff factor (i) for MX at this concentration.

Express your answer using two significant figures.

$i = 5,78$

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1. *Journal of the American Medical Association*, 2000; 283: 2639-2644.

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Mastering Chemistry Answer Key Chapter 1

Victor M. Corman



Mastering Chemistry Answer Key Chapter 1:

Decoding **Mastering Chemistry Answer Key Chapter 1**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Mastering Chemistry Answer Key Chapter 1**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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