

North Carolina

Standard Real Estate Exam Cheat Sheet



Important Links

North Carolina Real Estate Website

<https://www.ncrec.gov/>

North Carolina Real Estate Handbook

<https://candidate.psiexams.com/bulletin/>



Fees

- Examination Fee: \$56



Required Information to Make an Examination Reservation

- Your Authorization Notice
- Candidate Identification number listed on your authorization notice
- Your full name, address and a daytime telephone number
- The location of the test center you desire
- The name of the examination you will be taking
- Credit card number, check, or voucher



What to Bring

- Calculator that is allowed by your state testing board
- Proof that you've passed your class
- Valid photo ID
- Your testing permit which you should receive by mail or online before test
- Pen that writes in blue or black in case you have to fill out any forms
- Number 2 pencil and some blank scrap paper



Following Items are NOT allowed

- | | |
|-------------------|--|
| • cell phones | • PDAs |
| • purses | • calculators |
| • wallets | • cameras |
| • smart glasses | • video or digital recording devices |
| • keys | • watches |
| • backpacks | • pens |
| • briefcases | • lapel pins |
| • suitcases | • tie tacs |
| • food | • hats or caps |
| • drink | • weapons |
| • gum | • other items deemed inappropriate by proctors |
| • study materials | |
| • tablets | |
| • laptops | |

- 1. Contact A Friend in The Industry:**
Before you start your career in the industry, you should already have plenty of contacts. Remember, if someone has already taken the test, and passed, they can offer a lot of information.
- 2. Look at State Specific Testing Information:**
Different states have different testing requirements, so you should familiarize yourself with the rules of the state that you will be testing in.
- 3. Make an Appointment:**
It may sound weird to schedule your test before you start studying, but having a date scheduled will help you develop your study plan. Depending on your state, there may be dozens of testing locations, so having an appointment will let you know how long you have to study, and you can go ahead and make your travel arrangements.
- 4. Find Practice Exams:**
There are plenty of options when looking for practice exams, so you shouldn't struggle to find one that works for you. Try to find practice exams for your state and spend time testing yourself.
- 5. Take Notes:**
Once you start taking practice exams, you'll get a better understanding about the areas that you need to focus on even more. Take notes about the parts that you need some work on and focus your extra time on those topics.
- 6. Make Flashcards:**
Flashcards may remind you of studying for high school or college exams, but they're just as useful when you're prepping for your real estate exam.
- 7. Use Blue Ink:**
There are countless scientific studies that say that blue ink is better when you're trying to remember what's in your notes. According to research, blue calms the brain which helps you experience less stress when you're trying to study.
- 8. Get Rest Before the Exam:**
Again, depending on the state in which you're testing, you may have plenty of options when trying to find a testing location. Your mind will be better if you get plenty of rest the night before.

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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- \text{aq} + 2\text{Ag}^+ \text{aq}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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1 $\text{K}_3\text{PO}_4 \text{aq}$ $\text{Al}^{3+} \text{aq}$ $\text{AlPO}_4 \text{s}$ $3 \text{KNO}_3 \text{aq}$ $3 \text{H}^+ \text{aq}$ $\text{Al}^{3+} \text{aq}$ $3 \text{H}_2\text{O}$ aq $\text{AlPO}_4 \text{s}$ $3 \text{K}^+ \text{aq}$ $3 \text{PO}_4^{3-} \text{aq}$ $3 \text{H}^+ \text{aq}$ $3 \text{H}_2\text{O}$ aq 3mg s

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh₄ aq cr₂o₇²⁻ aq ag₂cr₂o₇ s 2nh₄ aq 2f aq because two nh₄ aq

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