

Name: _____

WRITING NEUTRAL FORMULAS FOR IONIC COMPOUNDS

Directions: Use a periodic table. Remember that A groups have the following ionic charges: Group 1 (+1), Group 2 (+2), Group 3 (+3), Group 5 (-3), Group 6 (-2), and Group 7 (-1). Transition metals can have variable positive charge, which is given to you in the compound name (ex. Iron (III) has a +3 charge whereas Copper (II) has a +2 charge).

#	Compound Name	Cation	Anion	Neutral Ionic Formula
1	Sodium chloride	Na ⁺	Cl ⁻	NaCl
2	Calcium fluoride	Ca ²⁺	F ⁻	CaF ₂
3	Aluminum chloride			
4	Aluminum oxide			
5	Beryllium fluoride			
6	Sodium oxide			
7	Copper (II) sulfide			
8	Aluminum fluoride			
9	Manganese (III) oxide			
10	Magnesium nitride			
11	Sodium sulfide			
12	Iron (III) oxide			

Formulae Of Ionic Compounds Answers

Léo Pomar

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Formulae Of Ionic Compounds Answers:

Formulae Of Ionic Compounds Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Formulae Of Ionic Compounds Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

<https://staging.conocer.cide.edu/files/scholarship/fetch.php/Gridiron%20Gourmet%20Chow%20For%20Champions.pdf>

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