

Self-Introduction

My name is Zhao Yan. I am studying at Nanjing Foreign Language School and now busy preparing for college study. Many years of hard working and strict training have made me a qualified middle school student with high qualities.

I have a good relationship with my classmates and teachers as I am generally sociable in personality. I am thoughtful and helpful in school. As for my study, I pay special attention to the development of my thinking manners so that I become quick in thinking and broad in knowledge. Seeking reality and progress, I have achieved excellence in my study. I particularly love such subjects as math, physics, biology and English. I am strong in logical thinking and oral expressing. I am creative and unique in views on certain matters. I have won many awards in the above subjects (see the copies of these certificates).

Morally I may say I am also excellent, because I am strict with myself, following the school's rules and regulations, taking active part in various activities, displaying great interest in our country's important events and world affairs. I love our motherland and our people and I am trying to get the party membership. All this has led me to win the "Xiang Yu Scholarship".

In addition, I have developed many hobbies in my life. For example, I love reading, computer and Internet, music and sports. I have been a champion several times in our school's sports meet.

In short, as a student studying at Nanjing Foreign Language School, I am determined to make my own contribution to our country's diplomatic affairs.

How To Write An Introduction Paper About Yourself

Yijin Wang



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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+ + \text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Ag}_2\text{Cr}_2\text{O}_7$

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web jan 29 2020 the ionic form of this reaction is $2\text{Na}^+ + 2\text{Cl}^- + \text{Cu}^{2+} + \text{SO}_4^{2-} \rightarrow 2\text{Na}^+ + \text{SO}_4^{2-} + \text{CuCl}_2$ the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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net ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g} \rightarrow \text{H}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ here is a video on spectator ions
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web the six NO_3^- ions and the six Na^+ ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2 \text{s}$

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+ \text{aq} + \text{OH}^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l}$ the spectator ions in $\text{KOH} \text{aq}$ and $\text{HNO}_3 \text{aq}$ are $\text{K}^+ \text{aq}$ and $\text{NO}_3^- \text{aq}$ how to identify and list the

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