

CHEMISTRY

FORM FIVE

FIRST TERM – 03.01.2007 to 09.03.2007 (10 weeks)

19.03.2007 to 25.05.2007 (10 weeks)

THEME
LEARNING AREA: INTERACTION BETWEEN CHEMICALS
: 1. RATE OF REACTION

No. of week/ Weeks	Learning Objectives	Suggested Learning Activities	Learning Outcomes	Thinking Skills	Scientific Skills	Scientific Attitudes And Noble Values	Remarks/ Vocabulary
1	1.1 Analysing rate of reaction	Discuss : (a) the meaning of rate of reaction, (b) some examples of fast reactions, (c) some examples of slow reactions. Discuss to identify observable changes to reactants or products and its method of measurement in order to determine the rate of reaction. Carry out an activity involving a reaction between zinc and acid, and plot a graph to determine average rate of reaction and the rate of reaction at any given time. Carry out problem solving activities involving rates of reaction.	A student is able to: - state what rate of reaction is, - identify observable changes to reactants or products for determining rate of reaction, - determine average rate of reaction, - determine the rate of reaction at any given time from a graph, - solve numerical problems involving average rate of reaction, - solve numerical problem involving rate of reaction at any given time.	- generation of ideas - relating making hypothesis - making conclusion - problem solving	- Observing - Hypothesising - Communicating - Interpreting data	- Being honest and accurate in recording and validating data. - Being cooperative - Being systematic	reactant- bahan tindak balas product- hasil tindak balas rate of reaction- kadar tindak balas observable change- perubahan yang dapat diperhatikan

1

No. of week/ Weeks	Learning Objectives	Suggested Learning Activities	Learning Outcomes	Thinking Skills	Scientific Skills	Scientific Attitudes And Noble Values	Remarks/ Vocabulary
2	1.2 Synthesising factors affecting the rate of reaction	Discuss possible factors affecting the rate of reaction. Design and carry out activities to investigate factors affecting the rate of reaction, i.e. size of reactant, concentration, temperature and catalyst. Some suggested reactions: (a) a reaction between calcium carbonate, CaCO_3, and hydrochloric acid, HCl, (b) a reaction between sodium thiosulphate, $\text{Na}_2\text{S}_2\text{O}_3$, and sulphuric acid, H_2SO_4, (c) decomposition of hydrogen peroxide, H_2O_2, in the presence of a catalyst. Use computer simulations to investigate how the movement and collision of particles in a reaction are affected by temperature, size of reactant, pressure, concentration and catalyst. Collect and interpret data to explain factors affecting the rate of reaction in the following: (a) combustion of charcoal, (b) storing food in a	A student is able to: - design experiment to investigate factors affecting the rate of reaction, - give examples of reactions that are affected by size of reactant, concentration, temperature and catalyst, - explain how each factor affects the rate of reaction, - describe how factors affecting the rate of	- generation of ideas - relating making hypothesis - analyzing - contributing - making conclusion - predicting making inference - problem solving	- experimenting - predicting - interpreting data - inferring	- being honest and accurate in recording and validating data - being systematic - being cooperative - being confident and independent	catalyst - mampin decomposition- penguraian

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Lei Shi



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Table of Contents Federal Scheme Of Work For Sschemistry

1. Understanding the eBook Federal Scheme Of Work For Sschemistry
 - The Rise of Digital Reading Federal Scheme Of Work For Sschemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Federal Scheme Of Work For Sschemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Federal Scheme Of Work For Sschemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Federal Scheme Of Work For Sschemistry
 - Personalized Recommendations

- Federal Scheme Of Work For Sschemistry User Reviews and Ratings
- Federal Scheme Of Work For Sschemistry and Bestseller Lists
- 5. Accessing Federal Scheme Of Work For Sschemistry Free and Paid eBooks
 - Federal Scheme Of Work For Sschemistry Public Domain eBooks
 - Federal Scheme Of Work For Sschemistry eBook Subscription Services
 - Federal Scheme Of Work For Sschemistry Budget-Friendly Options
- 6. Navigating Federal Scheme Of Work For Sschemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Federal Scheme Of Work For Sschemistry Compatibility with Devices
 - Federal Scheme Of Work For Sschemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Federal Scheme Of Work For Sschemistry
 - Highlighting and Note-Taking Federal Scheme Of Work For Sschemistry
 - Interactive Elements Federal Scheme Of Work For Sschemistry
- 8. Staying Engaged with Federal Scheme Of Work For Sschemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Federal Scheme Of Work For Sschemistry
- 9. Balancing eBooks and Physical Books Federal Scheme Of Work For Sschemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Federal Scheme Of Work For Sschemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Federal Scheme Of Work For Sschemistry
 - Setting Reading Goals Federal Scheme Of Work For Sschemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Federal Scheme Of Work For Sschemistry
 - Fact-Checking eBook Content of Federal Scheme Of Work For Sschemistry

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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