Chemistry EOC Review - Answer Key

- a. 68,200,000 cg
 b. 5000 mm
 c. 0.0125 kL
 d. 0.548 dm
- 2) -21.5 mL
- production of heat & light, formation of a gas, formation of a precipitate, color change
- 4) 33.64 g / 4.28 ml = 7.86 g/ml = Fe
- 5) m.p. -183°C = CH₄ (methane)
- 6) H₂ splint test (burning splint ignites hydrogen) O₂ – splint test (oxygen reignites glowing splint) H₂O – cobalt chloride paper (paper changes from blue to pink in the present of water vapor) CO₂ – splint test (carbon dioxide extinguishes burning splint);

limewater test (limewater turns cloudy when carbon dioxide is bubbled through it, producing CaCO₁)

- 7) The protons are <u>positively</u> charged particles found inside the <u>nucleus</u> along with the <u>neutrons</u>, which have no charge. The total number of these nucleons in an atom is the <u>mass # / atomic mass</u> of the atom. The electrons are the <u>negatively</u> charged particles found <u>orbiting in "clouds" around the nucleus</u>. Electrons can be gained by the atom to form <u>anions</u> or lost to form <u>cations</u>.
- 8) The thing that makes any element on the periodic table different from all other elements is the number of <u>protons</u> in its nucleus. <u>Isotopes</u> are atoms of the same type of element which have different <u>masses</u> because they have different numbers of <u>neutrons</u>.
- 9) a. ³⁶₁₉K *
 - b. ⁵⁷₂₆Fe
 c. ⁵³₁₆S²
- 105

chemical	atomic number	mass number	# of protons	# of electrons	# of neutrons	net charge
K.	19	42	19	18	2.3	+1
Sr	3-8	88	38	38	50	
F.	9	2:0	9	10	11	-3
P	1.5	31	15	15	1.6	0
Ca ²⁺	20	42	20	18	22	+2

- 11) uranium 238 238 92 = 146 neutrons
- 12) (1) 199.59g
 - (2) They contain the same # of protons (they are the same element), but different #s of neutrons (electrons don't matter, b/c they have virtually no mass), & thus different masses.

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J Rink

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language needed to succeed Through experimentation and research he discovered that for any subject there was a list of essential words that when mastered unlocked a student's ability to progress in the subject Lewis called this set of vocabulary the Insider's Words When he applied these Insider's Words the results were incredible His students began to learn with ease He was on his way to developing the landmark series of workbooks and applications to teach this Insider's Language to students around the world **Resources in Education** ,1996 Cracking the Virginia SOL Paul Foglino, Princeton Review (Firm), 2000 The Princeton Review realizes that acing the Chemistry exam is very different from getting straight As in school The Princeton Review doesn't try to teach students everything there is to know about chemistry only the techniques they ll need to score higher on the exam There's a big difference In Cracking the Virginia SOL EOC Chemistry TPR will teach test takers how to think like the test makers and Score higher by mastering what test takers need to know with proven techniques Use word association to solve problems when unsure of the correct answers Ace the exam by understanding the logic of the test makers and applying it to solutions This book includes 2 full length simulated exams All of TPR s sample test questions are just like the ones test takers will see on the actual End of Course Chemistry exam and TPR fully explains every solution Contents Include The Mystery Exams About the SOL Exams How to Crack the Chemistry EOC II Chemistry Review Atomic Structure Anatomy of the Periodic Table Chemical Compounds Chemical Reactions Chemistry Math Gases Phase Changes Solutions Equilibrium Acids and Bases How Chemical Reactions Happen Laboratory III The Princeton Review **Resources in Education** ,1995 American Laboratory ,2002 The United States Army and Practice Tests Navy Journal and Gazette of the Regular and Volunteer Forces ,1866 **Directory of Published Proceedings** ,1984

Growth Great Britain: Parliament: House of Lords: European Union Committee, 2006-03-16 This report by the European Union Committee examines the progress of the European Union in initiating a strategy for jobs and growth across the Community as a whole The background to this report stems from the Spring European Council meeting in 2000 in Lisbon and the launch of an economic reform agenda The Committee observes that since the Lisbon Agenda little progress has been made and the performance of many of the larger European economies has been poor The Committee has noted that certain worrying signs of protectionist behaviour have developed especially regarding barriers of cross border mergers The EU has recognized this weak performance and the Agenda was relaunched in 2005 with a greater focus on the key economic priorities of more growth and jobs Also all Member States are now required to produce an annual National Action Plan highlighting the policies being pursued to improve economic growth and increase employment The Committee sets out a number of recommendations to further push forward the priorities of growth and jobs including that the Commission should seek to complete the progress towards an internal market that Member States should influence one another in the development of good practice through statistical comparison of their economic progress and agree on quantifiable targets

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