

Name _____

Per _____

Chemistry Notebook

Mass

- Ability of a fluid to exert an _____ force on object immersed in a _____

- This usually happens in things that are _____ or filled with _____

Viscosity

- A fluid's resistance to _____ or how _____ or hard to _____ a fluid is

Density

Mass Per unit Volume Formula: Density = $\frac{\text{Mass}}{\text{Volume}}$

Physical Properties

- A property that can be measured without changing the substance

- Ex: _____

Chemical Properties

- A property that tells us how the substance _____ with other substances, the substance is _____ in the process. Ex: _____

Physical Change

- A change that does not change the identity of the substance involved. Ex: _____

Chemical Change

- A change that does change the identity of the substance involved. Bonds must be broken and new bonds formed.

- List five signs of chemical changes: _____

Ex: _____

Parts of the Periodic Table

Periodic Table of the Elements

Group	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Period 1	H	He																
Period 2	Li	Be	B	C	N	O	F	Ne										
Period 3	Na	Mg	Al	Si	P	S	Cl	Ar										
Period 4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Period 5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Period 6	Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Period 7	Fr	Ra	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	

Atomic number- equal to the number of _____ in atom of that element

Atomic mass- weighted _____ mass of all the _____ of an element

Mass number- _____ mass rounded to the nearest whole number,
number of _____ + number of _____

Metals: The elements _____ of the jagged line. They are usually good _____, are _____, and react to _____. All are _____, except mercury.

Shawna Lundberg

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