

*Honors Chemistry Midterm: Review Packet***PART 1**

1. Write in the number of significant figures for each of the following measurements:

3 3.00 mL      3 0.00290 g      4 50.00 m      2 0.070 kg      1 400 L

2. Using the following masses: 0.0562 kg, 124.213 g and 1635 mg.
- ← convert to grams*
- 
- The answer should be reported as:
- 
- g

*56.2g    124.213g    1.635g*

3. a) A 5.75 mL sample of mercury has a measured mass of 77.05 g. The density is
- 13.49/mL

$$D = \frac{m}{V} = \frac{77.05g}{5.75mL} =$$

- b) Mercury's accepted density is 13.53 g/mL. The % error in your measurement is:
- .96170

$$\frac{|\text{actual} - \text{exp}|}{\text{actual}} \times 100 \quad \frac{|13.53g/mL - 13.49g/mL|}{13.53g/mL} \times 100$$

4. In the diagram below:



Is this person accurate? no - not in center

Is this person precise? yes - close together

5. Identify the following as a chemical change (CC), physical change (PC), chemical property (CP), or physical property (PP):

a) burning wood CCb) mass of wood PPc) flammability of wood CPd) chopping up wood PC

# Honors Chemistry Student Learning Objective Review Packet Answers

**RC Schank**



## **Honors Chemistry Student Learning Objective Review Packet Answers:**

## **Honors Chemistry Student Learning Objective Review Packet Answers** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has be more evident than ever. They have the ability to inspire, provoke, and ignite change. Such could be the essence of the book **Honors Chemistry Student Learning Objective Review Packet Answers**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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