#### CH301 Worksheet I Answer Key: High School Chemistry Questions.

For each problem, identify the kind of question BEFORE answering it.

1. A 25 kg ball has a kinetic energy of 225 J. What is the speed of the ball?

What kind of question is this? energy

3 mm/s

How many molecules of water are there in 100 mL of water? (The density of water = 1.000 g/mL)
 What kind of question is this? stoichiometry

 $3.34 \times 10^{24}$  molecules

 Given that the atomic mass of carbon is given as 12.0107 g/mol, what are the approximate relative abundances of the two isotopes <sup>12</sup>C and <sup>13</sup>C? What kind of question is this? isotopes

98.93% 12C, 1.07% 13C

4. In the electrolysis of 2H<sub>2</sub>O → 2H<sub>2</sub> + O<sub>2</sub>, how many atom of O in O<sub>2</sub> are made from 4 g of H<sub>2</sub>O? What kind of question is this? stoichiometry

 $1.34 \times 10^{23}$  molecules

5. What are the most likely ionic forms of (a) sodium and (b) calcium?

What kind of question is this? periodic trends

(a) Na"; (b) Ca2"

What is the product of a reaction between metallic lithium and oxygen, O<sub>2</sub>?

What kind of question is this? oxidation numbers

Li<sub>2</sub>O

Name the following species: (a) Cl', (b) NO<sub>3</sub>', (c) C<sub>3</sub>H<sub>8</sub>. What kind of question is this? naming

(a) chloride; (b) nitrate; (c) propane

The combustion of ethanol has the following equation: C<sub>2</sub>H<sub>2</sub>OH + 3 O<sub>2</sub> → 2 CO<sub>2</sub> + 2 H<sub>2</sub>O
 What mass of CO<sub>2</sub> is formed from 1 kg of ethanol? What kind of question is this? stoichiometry

 $1.913 \, \mathrm{kg}$ 

Vitamin K contains 78.95% C, 3.95% H, and 21.05% O by mass. What is its empirical formula?
 What kind of question is this? empirical formula calculation

 $C_5H_5O$ 

10. Given that vitamin K (see question 9) has molecular weight of 158.15 g/mol, what is its molecular formula? What kind of question is this? molecular formula calculation

 $C_{10}H_0O_2$ 

# **High School Chemistry Test Questions And Answers**

Xiaolong Qi

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