# MAY/JUNE SENIOR CERTIFICATE (SC)/NATIONAL SENIOR CERTIFICATE (NSC) EXAMINATIONS TIMETABLE 2024 -FINAL

| WEEK 1              | 09:00  | 14:00  |
|---------------------|--|--|
| Monday<br>29 April  | Hindi, Gujarati, Tamil, Telegu, Urdu HL P1 (2hrs)<br>FAL P1 (2hrs), SAL P1 (2hrs)<br>Arabic, French, Hebrew, Italian, Mandarin, Modern<br>Greek, Serbian, Spanish SAL P1 (2hrs)<br>Latin SAL P1 (3hrs)     | English HL P1 (2hrs)*<br>English FAL P1 (2hrs)*<br>English SAL P1 (2hrs)                 |
| Tuesday<br>30 April | Portuguese, German HL P1 (2hrs)<br>Portuguese, German FAL P1 (2hrs)<br>Portuguese, German SAL P1 (2hrs)  | Religion Studies P1 (2hrs)*<br>Information Technology P2 Theory (3hrs)*                  |
| Wednesday<br>1 May  |  |  |
| Thursday<br>2 May   | Hindi, Gujarati, Tamil, Telegu, Urdu HL P2<br>(2½hrs)<br>FAL P2 (2hrs), SAL P2 (2hrs)<br>Arabic, French, Hebrew, Italian, Mandarin, Modern<br>Greek, Serbian, Spanish SAL P2 (2hrs)<br>Latin SAL P2 (2hrs) | Computer Applications Tech P2 Theory (3hrs)*   |
| Friday<br>3 May     | Portuguese, German HL P2 (2½hrs) Portuguese, German FAL P2 (2hrs) Portuguese, German SAL P2 (2hrs)   | Mathematics P1 (3hrs)* Mathematical Literacy P1 (3hrs)* Technical Mathematics P1 (3hrs)  |
| WEEK 2              | 09:00  | 14:00  |
| Monday<br>6 May     | Information Technology P1 (3hrs) Practical*  | Mathematics P2 (3hrs)* Mathematical Literacy P2 (3hrs) * Technical Mathematics P2 (3hrs) |
| Tuesday<br>7 May    | Engineering Graphics and Design P1 (3hrs)  | Accounting P1 (2hrs)*  |
| Wednesday<br>8 May  | Hindi, Gujarati, Tamil, Telegu, Urdu HL P3<br>(2½hrs)<br>FAL P3 (2½hrs)<br>Portuguese, German HL P3 (2½hrs)<br>Portuguese, German FAL P3 (2½hrs)   | Geography (Climate and Weather, Geomorphology and<br>Map Work) P1 (3hrs)*                |
| Thursday<br>9 May   | Sport and Exercise Science (3hrs)  | Afrikaans HL P1 (2hrs)*<br>Afrikaans FAL P1 (2hrs)*<br>Afrikaans SAL P1 (2hrs)           |
| Friday<br>10 May    | Engineering Graphics and Design P2 (3hrs)  | Physical Sciences (Physics) P1 (3hrs)*<br>Technical Sciences P1 (3hrs)                   |

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web solution v1 2 50 l v2 t1 273 30 303 k t2 273 70 343 k we will be applying charles law as pressure is constant here v1 t1 v2 t2 or v2 v1 t1 x t2 2 50 l 303 k x 343 k 2 83 l at 90 c a helium sample has a volume of 500 ml determine the temperature at which the volume of the liquid will become 240 ml