

# MASS MOLE

## STOICHIOMETRY WORKSHEET

1. Consider the following reaction:



- If 15 g of  $\text{N}_2\text{O}_4$  was produced, how many moles of  $\text{O}_2$  were required?
- If  $4 \times 10^{-2}$  moles of oxygen reacted, how many grams of  $\text{N}_2$  were needed?

2. Consider the following reaction:



- What is the mass of potassium nitrate produced when 2.04 moles of potassium phosphate react?
- If 5.8 g of aluminum phosphate are formed, how many moles of aluminum nitrate reacted?

3. Consider the following reaction:



- If you have 5.5 mol of  $\text{CaC}_2$ , how much  $\text{C}_2\text{H}_2$  do you get?
- How many moles of water are needed when 65 g of  $\text{CaC}_2$  have reacted?

4. Hydrogen sulfide gas burns in oxygen to produce sulfur dioxide and water vapor:



What mass of oxygen gas is consumed in a reaction that produces 4.6 mol  $\text{SO}_2$ ?

# Mass Mass Stoichiometry Packet Answers

**David Kirk**



## **Mass Mass Stoichiometry Packet Answers:**

**The Practice of Chemistry Study Guide & Solutions Manual** Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises      **Study Guide to Accompany Calculus for the Management, Life, and Social Sciences** Clyde Metz, 1984-01-01 Study Guide to Accompany Calculus for the Management Life and Social Sciences      **STOICHIOMETRY** NARAYAN CHANGDER, 2024-04-01 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquiziz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today's academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests quizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also provides you with the skills and information you need to achieve a good score in challenging tests or competitive examinations Whether you have studied the subject on your own read for pleasure or completed coursework it will assess your knowledge and prepare you for competitive exams quizzes trivia and more

General Chemistry Workbook Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material      **Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane  $\text{CH}_4$  from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with

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