

# Equilibrium Lab

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*Instructions: Complete all sections of this report and answer all questions in complete sentences for full credit.*

**Title:** Equilibrium Lab Report

**Objective(s):**

**Hypothesis:** Complete the following statements for your hypotheses. Choose one of the options in brackets by highlighting, bolding, or placing your answer in a different color.

If the amount of reactant increases, the rate of the forward reaction will [**increase** or decrease] and the rate of the reverse reaction will [increase or **decrease**].

If the amount of reactant decreases, the rate of the forward reaction will [increase or **decrease**] and the rate of the reverse reaction will [**increase** or decrease].

If the amount of product increases, the rate of the forward reaction will [increase or **decrease**] and the rate of the reverse reaction will [**increase** or decrease].

If the amount of product decreases, the rate of the forward reaction will [**increase** or decrease] and the rate of the reverse reaction will [increase or **decrease**].

**Materials:** For this lab, you are not expected to list the materials. But still notice the materials (lab equipment and chemicals) used in the lab to help you analyze the data, write the conclusion, and answer the post-lab questions.

**Variables:** List variables below as you work through the simulation.

Type	Variables
Controlled	Temperature, pressure, surface pressure
Independent	the concentration of reactants and the concentration of the products
Dependent	the rate of both the forward and reverse reactions

**Procedure:** Follow the steps described in the virtual lab. Record your observations and interpretations below. Equilibrium reaction in the lab:



**Data:** For each trial, do the following:

# Lab 11 Rate And Equilibrium Report Answer

**L Manion**



## Lab 11 Rate And Equilibrium Report Answer:

*General, Organic, and Biological Chemistry Study Guide and Selected Solutions* Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems

**Publications, Reports, and Papers for 1966 from Oak Ridge National Laboratory** F. M. O'Hara, Ann S. Klein, 1967

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**Scientific and Technical Aerospace Reports**, 1981-04

Nuclear Science Abstracts, 1975

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U.S. Government Research & Development Reports, 1970

Energy Research Abstracts, 1986

**Atomic Energy Research Reports for Sale by the U.S. Department of Commerce, Office of Technical Services** Business and Defense Services Administration, 1964

**TID.**, 19??

Bibliography of Scientific and Industrial Reports, 1965-07

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Reactor Technology, 1967

Reactor and Fuel-processing Technology, 1967

**Power Reactor Technology**, 1965

**Government Reports Announcements & Index**, 1981-05

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