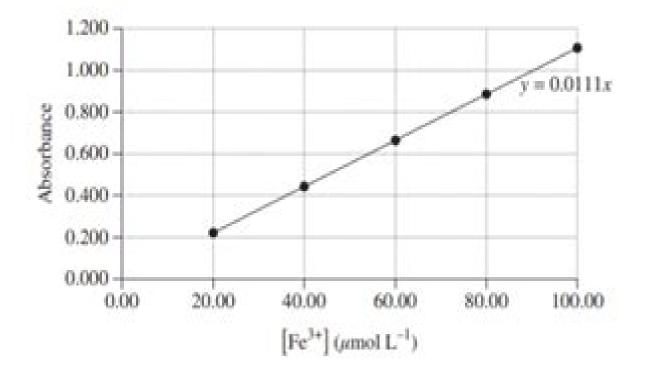
A colorimeter was used to calculate the percentage of iron in a 0.200 gram tablet. The tablet was dissolved and oxidised, then reacted with thiosulfate according to the equation

$$Fe^{3+}(aq) + SCN^{-}(aq) \rightarrow [FeSCN]^{2+}(aq)$$
.

The resulting solution was made up to 200 mL with distilled water. The absorbance of the final solution was measured to be 0.6105.

The calibration curve below shows the absorbance of various concentrations of Fe3\*.



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**R Barnett** 

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